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# 1 Development of pore water pressure during freeze-thaw cycles in 2 weather conditions

3  
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8  
9 **Abstract.** To assess the risk of frost damage to building materials, the mechanisms of pressure  
10 development in a material due to freezing under weather conditions should be adequately understood.  
11 In this study, we have demonstrated the evolution of pore water pressure in a brick wall during cyclic  
12 freezing and thawing caused by changes in air temperature and solar radiation. We developed a  
13 coupled hygrothermal and mechanical model of brick walls based on the theory of poromechanics.  
14 The calculation results showed that the water pressure in a south facing wall increases significantly  
15 during the night in a cold region. For a south facing wall, the surface thaws during the day due to solar  
16 radiation and freezes again at night, while the interior remains frozen throughout the day. We conclude  
17 that such incomplete freeze-thaw cycles in the weather conditions are much more damaging to building  
18 materials than a monotonous repetition of complete freezing and thawing.

19  
20 **Keywords:** Frost damage, masonry, numerical simulation, weather condition, solar radiation,  
21 poromechanics

## 22 23 **1. Introduction**

24 Assessing the risk of deterioration of buildings and structures is of great importance in achieving  
25 sustainable development. Frost action is one of the main causes of such deterioration, and considerable  
26 efforts have been made to assess the risks of frost damage associated with weather conditions. Such

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27 studies have been conducted mainly using various indexes reflecting environmental conditions, such  
28 as the number of freeze-thaw cycles, the length of time the material is frozen, or the air temperature  
29 (Grossi, et al., 2007). Hama et al. (1999) and Noguchi and Hama (2021) related environmental factors  
30 to the number of freeze-thaw cycles specified in ASTM C666 A method (2015) to predict durability  
31 of a material under weather conditions based on the freeze-thaw test results. In addition, damage  
32 functions such as the modified version of the winter index (MWI), time of frost (TOF), amount of  
33 water (ASW), and indicative freeze-thaw cycles (IFTC) were used with hygrothermal simulation tools  
34 to compare the risks of frost damage to building walls with different envelope types or in different  
35 climates (Kočí, et al., 2017; Kočí, et al., 2018). Zhou et al. (2017) proposed the FTDR index based on  
36 the amplitude of the ice saturation degree of a building wall. This index is calculated using a  
37 hygrothermal simulation that includes freezing and thawing processes, and its concept reflects the  
38 characteristics of freeze-thaw cycles in a real environment, where freezing often resumes before  
39 thawing is complete.

40 These approaches are well suited for comparing the risks of building deterioration under different  
41 climatic conditions or with different wall structures. At present however, the mechanisms of frost  
42 damage under weather conditions are not well understood, and there appears to be a variety of  
43 phenomena that need to be uncovered for a more reliable assessment. For one of these phenomena,  
44 Nakamura et al. (2011a; 2011b; 2016) proposed “blockage-type” freeze-thaw cycles, in which cyclic  
45 freezing and thawing creates an unfrozen area surrounded by a frozen area inside a material. The  
46 frozen area traps the unfrozen water inside and results in destructive pressure associated with volume  
47 increase due to freezing. This mechanism is similar to the closed container theory (Fagerlund, 1997),  
48 which describes the increase in pressure in a low permeability material that prevents unfrozen water  
49 from escaping from the material as the water volume increases due to freezing. In contrast, freezing  
50 originating from the material surface blocks the movement of unfrozen water from the inside in the  
51 “blockage-type” freeze-thaw cycles, rather than the low permeability of the material.

52 Similar effects blocking water movement in a material have been observed in freeze-thaw tests.  
53 Wardeh and Perrin (2008) and Sánchez de Rojas et al. (2011) discussed the effects of a wet cloth

54 placed on a roof tile specimen on the results of freeze-thaw tests using method E specified in the  
55 European standard (DIN EN 539-2, 2013) by comparing tests with and without a wet cloth <sup>1</sup>. The  
56 explanation is as follows. Because the wet cloth contains a large amount of water, a specimen surface  
57 in contact with the cloth freezes first during a cooling period and blocks the movement of water toward  
58 it, which increases the internal pressure and leads to greater stress and damage. In fired clay materials  
59 such as roof tiles and bricks, the likelihood of "blocking" type freeze-thaw cycles appears to be high  
60 because the freezing point depression of the contained water in these materials is low due to the large  
61 pore diameters and freezing progresses rapidly slightly below 0 °C, resulting in severe blocking of the  
62 movement of unfrozen water. However, the significance of the "blockage-type" freeze-thaw cycles  
63 for pressure development in real buildings or structures has not yet been clarified because it is difficult  
64 to adequately reflect real weather conditions in laboratory experiments.

65 Therefore, in this study, we aim to demonstrate the pressure evolution of trapped pore water during  
66 cyclic freezing and thawing under real weather conditions. To decipher the complex interaction  
67 between heat and moisture transfer and pressure change in a material induced by environmental  
68 condition changes, we used coupled hygrothermal and mechanical simulations that consider freezing  
69 and thawing processes based on the theory of poromechanics (Coussy, 2004; Coussy, 2005; Coussy &  
70 Monteiro, 2008). The rest of this study is organized as follows. In the second section, we summarize  
71 the studies by Nakamura et al. (2011a; 2011b; 2016) to clarify the environmental conditions that lead  
72 to degradation by "blockage-type" freeze-thaw cycles. In the third section, a coupled hygrothermal  
73 and mechanical model is developed, and its validity is confirmed by comparing the calculation results  
74 with the results of strain measurements on a brick specimen. Finally, in the fourth section, calculations  
75 are performed to demonstrate the effects of water blockage. We performed two calculations in which  
76 a material was subjected to multiple freeze-thaw cycles. In the first calculation, a material was exposed  
77 to a relatively simple temperature variations of the outside air. In contrast, real meteorological data  
78 (air temperature and solar radiation) in a cold region were input into the second calculation.

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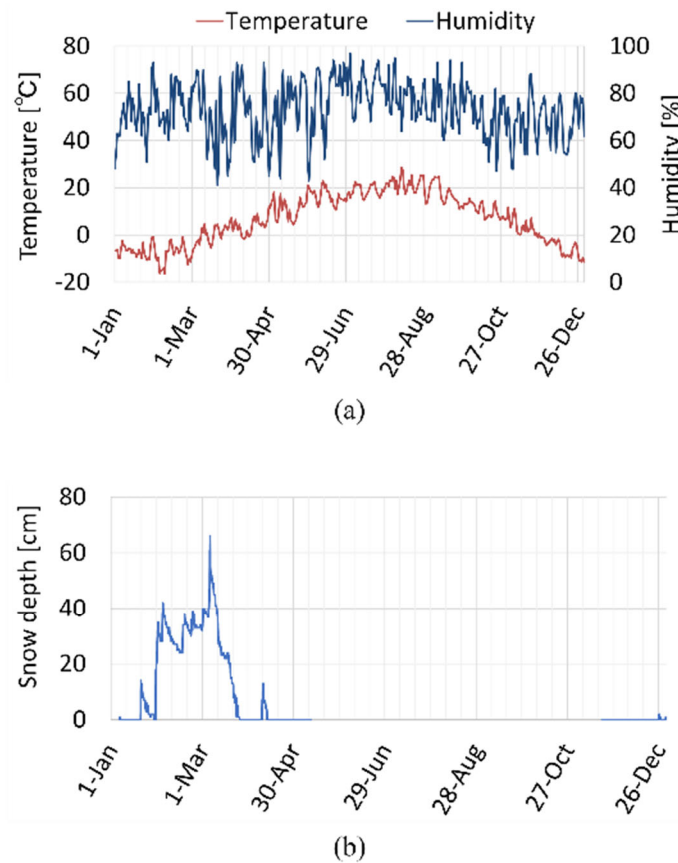
<sup>1</sup> Wardeh & Perrin (2008) compared tests with and without wet cloth using the temperature conditions specified in the E method of EN 539-2. Sanchez de Rojas et al (2011) compared experiments using the C and E methods of EN 539-2.

79

## 80 2. Observation of the deterioration of brick walls in the literature

81 Nakamura et al. (2011a; 2016) reported the deterioration of brick walls in Kitami City, Hokkaido  
82 Prefecture, Japan. Fig. 1 shows the daily average temperature, humidity, and hourly snow depth in the  
83 region in 2020 provided by the Japan Meteorological Agency (2021). Humidity data in Kitami City  
84 are not provided. Instead, data are shown for the neighboring city of Abashiri, which is about 40 km  
85 from Kitami City. The region is classified as a humid continental climate zone according to the Köppen  
86 climate classification (Kottek, et al., 2006), and is characterized by low temperatures and continuous  
87 snow cover in winter.

88



89

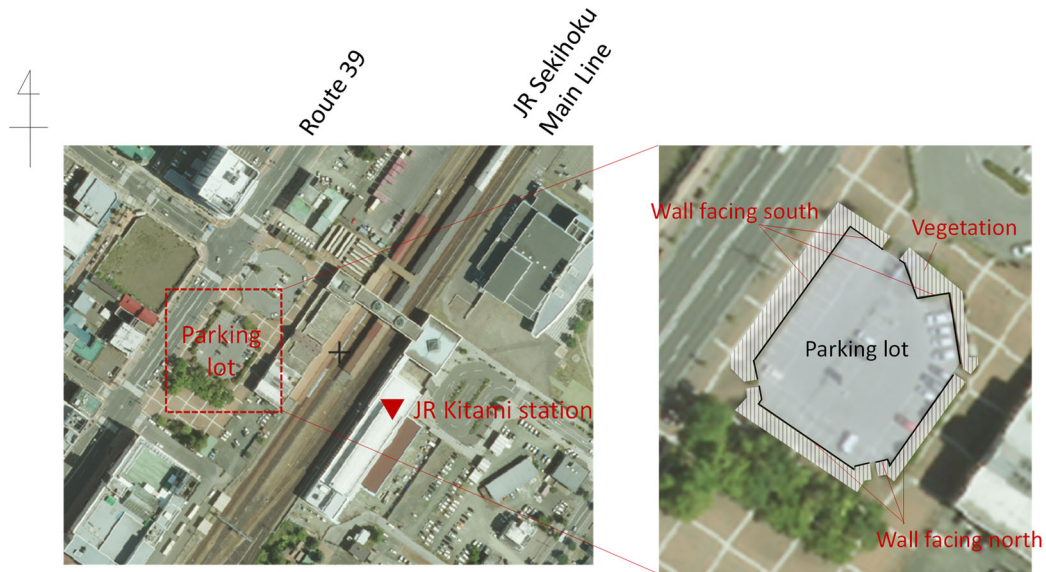
90 **Fig. 1.** (a) Daily average temperature and humidity and (b) hourly snow depth in the region including  
91 Kitami City.

92

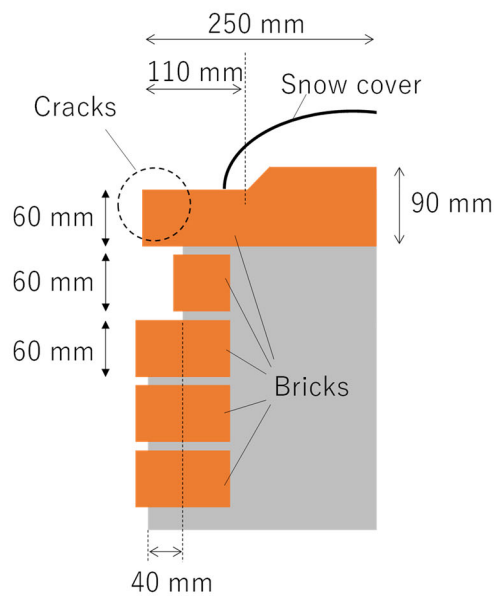
93       The brick walls are 130 cm high and are located near the Japan Railway (JR) Kitami Station. Fig.  
94 2 (a) shows an aerial photograph of the walls and station taken in 2015 (Geospatial Information  
95 Authority of Japan, 2021). The walls surround a parking lot and, therefore, are oriented in different  
96 directions. The observations of Nakamura et al. (2011a) can be summarized as follows:

- 97 • There were many cracks, including fatal ones, and flaking on the south facing walls, whereas tiny  
98 cracks were observed on the north facing walls.
- 99 • Most of the damage was observed on the bricks at the top of the walls, as shown in Fig. 2 (b),  
100 where the bricks were most exposed to the external environment.
- 101 • They conducted measurements of the temperatures of outside air and the on the surface and  
102 outside temperature of the brick walls. Based on the measurement results in late winter, the  
103 surface temperature of the south facing wall can reach 30 °C due to solar radiation. Therefore, it  
104 is assumed that the bricks are subject to a cycle of thawing during the day and freezing at night.

105



(a)



(b)

106

107 **Fig. 2** (a) Aerial photograph of the parking lot with brick walls (Geospatial Information Authority of

108 Japan, 2021) (retouched by the authors) and (b) schematic cross-section of the brick wall (prepared by

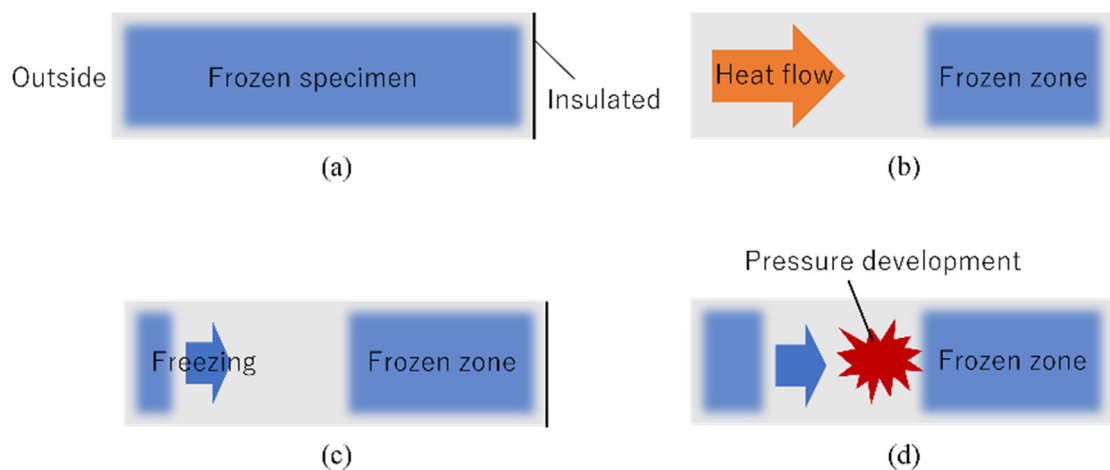
109 the authors with reference to Nakamura, et al. (2011a)).

110

111 Nakamura et al. (2011b; 2016) also conducted laboratory experiments to investigate the causes of

112 the deterioration. In their experiment, brick specimens were exposed to either of two patterns of freeze-

113 thaw cycles. While the first experiment was an ordinary one-dimensional freeze-thaw test, in the  
 114 second experiment they used a special temperature condition to simulate freeze-thaw cycles under  
 115 real weather conditions in a cold region. The assumption was that the bricks began to thaw from  
 116 the surface during the day due to solar radiation, but the interior remained frozen, resulting in a  
 117 significant increase in the water pressure as the unfrozen water became trapped by frozen zones when  
 118 freezing resumed on the surface the next night. Fig. 3 shows schematic representations of the freeze-  
 119 thaw processes during the second experiment. At the end of the cooling period, the specimens were  
 120 completely frozen (Fig. 3 (a)) and gradually thawed at the surface during the heating period (Fig. 3  
 121 (b)). However, the next cooling period started before the specimen was completely thawed (Fig. 3 (c)),  
 122 and an unfrozen zone surrounded by frozen zones formed inside (Fig. 3 (d)). In their experiment, the  
 123 minimum temperature was set at  $-8$ ,  $-15$ , and  $-25$  °C, and the maximum temperature was set at  $20 \pm$   
 124  $5$  °C. The second cooling period started when the temperature became positive at a distance of 10 mm  
 125 or 60 mm from the surface. They showed that cracks appeared only after the second freezing in the  
 126 second experiment, while the one-dimensional freeze-thaw test did not cause such deterioration.  
 127 Therefore, they concluded that just one “blockage-type” freeze can be very damaging to the materials.  
 128



129  
 130 **Fig. 3.** Schematic representation of the freezing and thawing processes during the “blockage-type”  
 131 freeze-thaw experiment performed by Nakamura et al. (2011b; 2016).

132

133 Although Nakamura et al. (2011a; 2016) did not observe building envelopes, the brick walls were  
134 exposed to very common environmental conditions in cold regions, and there were many similarities  
135 in boundary conditions with building envelopes. In addition, Takami and Ishizaki (2012) observed that  
136 a building brick wall was more seriously damaged by frost action in the same climate zone (Otaru,  
137 Hokkaido Prefecture, Japan) when exposed to sunlight. Therefore, it is very likely that the building  
138 walls are damaged by mechanisms similar to those studied by Nakamura et al. (2011a; 2016)

139

### 140 3. Coupled hygrothermal and mechanical model

#### 141 3.1. Fundamental equations

##### 142 3.1.1. Constitutive equations

143 In this study, we have assumed that the materials are nearly saturated with liquid water and ice due  
144 to the continuous water supply from the snow cover. Plastic deformation of a material is not considered  
145 in the calculations. We used the following constitutive equations for saturated and frozen materials  
146 based on the theory of poromechanics (Coussy, 2004; Coussy, 2005; Coussy & Monteiro, 2008):

147

$$148 \quad \boldsymbol{\sigma} = \mathbf{D}[\boldsymbol{\varepsilon} - \alpha(T - T_r)\mathbf{I}] - b(S_i P_i + S_l P_l)\mathbf{I} \quad (1)$$

$$149 \quad \varphi_i = b\varepsilon_{vol}S_i + \frac{P_i}{N_{ii}} + \frac{P_l}{N_{il}} - a_i(T - T_r) \quad \varphi_l = b\varepsilon_{vol}S_l + \frac{P_i}{N_{li}} + \frac{P_l}{N_{ll}} - a_l(T - T_r) \quad (2)$$

150

151 where  $a_j$  is a coefficient related to the volumetric thermal expansion of the pore volume occupied by  
152 phase  $j$  ( $j = i$  or  $l$ ) [ $\text{K}^{-1}$ ];  $b$  is the Biot coefficient;  $N_{ii}$ ,  $N_{il}$ ,  $N_{li}$  and  $N_{ll}$  are the generalized Biot coupling  
153 moduli [Pa];  $P$  is the pressure [Pa];  $S$  is the saturation;  $T$  is the temperature [K];  $T_r$  is the reference  
154 temperature [K];  $\alpha$  is the thermal expansion coefficient of the skeleton [ $\text{K}^{-1}$ ];  $\varepsilon_{vol}$  is the volumetric  
155 strain;  $\varphi_j$  is the increment of porosity occupied by phase  $j$  ( $j = i$  or  $l$ );  $\mathbf{D}$  is the elastic stiffness tensor  
156 of the material [Pa];  $\mathbf{I}$  is the second order unit tensor,  $\boldsymbol{\varepsilon}$  is the strain tensor; and  $\boldsymbol{\sigma}$  is the total stress  
157 tensor [Pa]. The subscripts  $i$  and  $l$  represent the ice and liquid water, respectively. Eqs. (1) and (2)

158 represent the deformation of the material and pore space, respectively due to the pressure changes of  
 159 the liquid and ice phases as well as the thermal expansion and external force. Unlike our previous  
 160 study (Fukui et al., 2023a), the model did not consider the anisotropy of the material properties, for  
 161 simplicity. Therefore, the thermoporoelastic material properties can be derived using the following  
 162 equation (Coussy, 2005; Coussy & Monteiro, 2008):

163

$$164 \quad b = 1 - \frac{K}{K_s} \quad (3)$$

$$165 \quad \frac{1}{N_{ii}} + \frac{1}{N_{il}} = \frac{b - \phi_0}{K_s} S_i \quad \frac{1}{N_{li}} + \frac{1}{N_{ll}} = \frac{b - \phi_0}{K_s} S_l \quad (4)$$

$$166 \quad a_j = 3\alpha(b - \phi_0)S_j \quad j = i, l \quad (5)$$

167

168 where  $K$  and  $K_s$  are the bulk moduli of the bulk material and its solid phase, respectively, and  $\phi_0$  is the  
 169 porosity in the reference state. According to Aichi and Tokunaga (2011), the following equations were  
 170 used to calculate  $N_{ii}$ ,  $N_{il}$ ,  $N_{li}$  and  $N_{ll}$ :

171

$$172 \quad \frac{1}{N_{ii}} = \frac{b - \phi_0}{K_s} S_i^2 \quad \frac{1}{N_{il}} = \frac{1}{N_{li}} = \frac{b - \phi_0}{K_s} S_j S_l \quad \frac{1}{N_{ll}} = \frac{b - \phi_0}{K_s} S_l^2 \quad (6)$$

173

174 As  $b$ ,  $K_s$ ,  $\alpha$ , and  $\phi_0$  are constants,  $S_j$  is only a variable to derive the poroelastic parameters  $b$ ,  $N_{ii}$ ,  $N_{il}$ ,  
 175  $N_{li}$ ,  $N_{ll}$ , and  $a_j$ .  $S_j$  is time and space dependent and determined based on the relationship described in  
 176 subsection 3.1.2. Using  $S_j$  for each time step and element, the poroelastic parameters were determined.

177

### 178 3.1.2. Equilibrium relationships between ice and liquid water

179 Assuming thermodynamic equilibrium conditions between the ice and liquid water in the pores, the  
 180 following Clausius-Clapeyron equation was used (Coussy & Monteiro, 2009):

181

182 
$$\frac{P_l}{\rho_l} - \frac{P_i}{\rho_i} = \frac{\Delta s}{\rho_i} (T - T_m) \quad (7)$$

183

184 where  $T_m$  is the melting point of bulk water [K],  $\Delta s$  is the melting entropy (= 1.2 [MPa/K]), and  $\rho$  is  
 185 the density [kg/m<sup>3</sup>]. The dependence of the density of water on pressure and temperature is expressed  
 186 by the following linearized form (Coussy, 2005; Coussy & Monteiro, 2008):

187

188 
$$\frac{1}{\rho_j} = \frac{1}{\rho_j^0} \left( 1 - \frac{P_j}{K_j} + 3\alpha_j (T - T_r) \right) \quad j = i, l \quad (8)$$

189

190 where  $\rho_j^0$  is the density of phase  $j$  ( $j = i$  or  $l$ ) in the initial state [kg/m<sup>3</sup>].

191 To express the relationship between pore water pressure and saturation, the well-known form  
 192 proposed by van Genuchten (1980) was applied to the freezing and thawing processes, taking into  
 193 account the difference in interfacial energy (Zeng, et al., 2011):

194

195 
$$S_l = \left[ \frac{1}{1 + \left( \beta \frac{\gamma_{al}}{\gamma_{il}} (P_i - P_l) \right)^n} \right]^m \quad (9)$$

196

197 where  $n$ ,  $m$ , and  $\beta$  are constants, and  $\gamma_{al}$  and  $\gamma_{il}$  are the interfacial energies between air and liquid water  
 198 (=  $73 \times 10^{-3}$  [J/m<sup>2</sup>] at 20 °C) and between ice and liquid water (=  $33 \times 10^{-3}$  [J/m<sup>2</sup>] at 0 °C), respectively.

199

### 200 3.1.3. Conservation equations

201 The conservation equations used in this study are the same as those used in our previous study (Fukui  
 202 et al., 2023a), but the anisotropy of the material properties was ignored for simplicity. The momentum  
 203 balance is expressed as follows:

204

205 
$$\nabla \cdot \boldsymbol{\sigma} + \mathbf{F} = 0 \quad (10)$$

206

207 where  $\mathbf{F}$  denotes the body force vector [N/m<sup>3</sup>]. Assuming that there are no external sources of force,  
 208  $\mathbf{F} = 0$ . The equations for heat and moisture conservation are expressed as follows:

209

210 
$$\frac{\partial}{\partial t}(CT - HM_i) = \nabla \cdot (\lambda \nabla T) \quad (11)$$

211 
$$\frac{\partial}{\partial t}(M_i + M_l) = \nabla \cdot (\lambda'_i \nabla P_i) \quad (12)$$

212

213 where

214

215 
$$C = c_d \rho_d + c_i M_i + c_l M_l \quad (13)$$

216 
$$M_j = \rho_j (\phi_0 S_j + \varphi_j) \quad j = i, l \quad (14)$$

217

218 where  $c$  is the specific heat [J/(kg·K)],  $H$  is the latent heat of solidification ( $334 \times 10^3$  [J/kg]),  $\lambda$  is the  
 219 thermal conductivity [W/(m·K)], and  $\lambda'_i$  is the moisture permeability [kg/(m·s·Pa)]. Subscript  $d$   
 220 represents the bulk material in the dry state.  $C$  is the volumetric specific heat of the wet material  
 221 [J/(m<sup>3</sup>·K)] and  $M_j$  is the mass of phase  $j$  contained in a unit volume of the bulk material [kg/m<sup>3</sup>]. When  
 222 water freezes, the volume increases (which is considered as the density change in the model), and the  
 223 material deforms in conjunction with an increase in liquid and ice pressures to satisfy the mass balance  
 224 (Eq. (14)) and the constitutive equations (Eqs. (1) and (2)). In addition, the unfrozen water flows out  
 225 of the freezing area due to the pressure gradient according to Eq. (14), and the importance of the  
 226 pressure depends on the moisture permeability.

227

228 **3.2. Numerical solution**

229 The calculations were performed in two dimensions assuming a plane-strain state. The conservation  
 230 equations were spatially discretized using Galerkin's finite-element method.  $T$ ,  $P_l$  and the  
 231 displacement vector  $\mathbf{u}$  [m] were chosen as the main variables and expressed by vectors of nodal values  
 232  $\mathbf{T}_e$ ,  $\mathbf{P}_e$ , and  $\mathbf{u}_e$ , and linear shape functions  $\mathbf{N}$  and  $\mathbf{N}_u$ :

233

$$234 \quad T = \mathbf{N}\mathbf{T}_e \quad P_l = \mathbf{N}\mathbf{P}_e \quad \mathbf{u} = \mathbf{N}_u\mathbf{u}_e \quad (15)$$

235

236 After discretization, we obtain the following system of equations:

237

$$238 \quad \begin{bmatrix} \mathbf{C}_{TT} & \mathbf{C}_{TP} & \mathbf{C}_{Tu} \\ \mathbf{C}_{PT} & \mathbf{C}_{PP} & \mathbf{C}_{Pu} \\ \mathbf{C}_{uT} & \mathbf{C}_{up} & \mathbf{C}_{uu} \end{bmatrix} \frac{\partial}{\partial t} \begin{bmatrix} \mathbf{T}_e \\ \mathbf{P}_e \\ \mathbf{u}_e \end{bmatrix} \quad (16)$$

$$= \begin{bmatrix} \mathbf{K}_{TT} & \mathbf{0} & \mathbf{0} \\ \mathbf{0} & \mathbf{K}_{uu} & \mathbf{0} \\ \mathbf{0} & \mathbf{0} & \mathbf{0} \end{bmatrix} \begin{bmatrix} \mathbf{T}_e \\ \mathbf{P}_e \\ \mathbf{u}_e \end{bmatrix} + \begin{bmatrix} \mathbf{f}_T \\ \mathbf{f}_P \\ \partial\mathbf{f}_u/\partial t \end{bmatrix}$$

239

240 where Eq. (10) is differentiated by time after substituting Eq. (1). The components  $\mathbf{C}$  and  $\mathbf{K}$  were  
 241 described in our previous study (Fukui et al., 2023a). The finite difference method was used for the  
 242 discretization in time, and the fully implicit method was used.

243

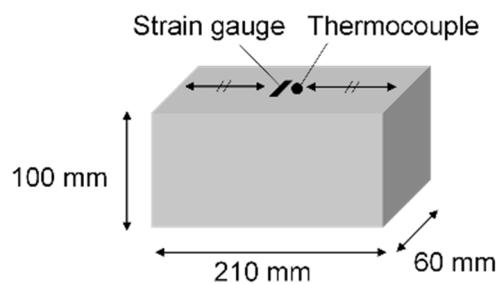
244 **3.3. Model validation**

245 *3.3.1. Strain measurement*

246 Strain measurement results reported in the previous study (Fukui et al., 2023b) were used for model  
 247 validation. The measurement was performed under simple boundary conditions and was suitable to  
 248 capture the deformation caused by the pressure evolution due to freezing and thawing. No constraining  
 249 force was applied to the specimen. The measurement method is as follows. A brick specimen  
 250 manufactured in Aichi Prefecture, Japan, was used. The bricks had dimensions of 210 mm × 100 mm  
 251 × 60 mm. To reduce the three-dimensional effects of heat and moisture transfer in the specimen as

252 much as possible, the measurements were made in the center of the 210 mm long sides, as shown in  
253 Fig. 4. A strain gage (KFLB-5-120-C1-11 R3M3, Kyowa Electronic Instruments Co., Ltd.) was placed  
254 in the center of the 210 mm × 60 mm top surface and a thermocouple was attached nearby with a  
255 cyanoacrylate adhesive. The direction in which the strain was measured is the same as the height  
256 direction of the brick wall shown in Fig. 2 (a). Strain and temperature were also measured on one of  
257 the 210 mm × 100 mm side surfaces (not shown in the figure). The specimen was vacuum-saturated  
258 and covered with polyvinylidene chloride wrap to prevent evaporation from the surfaces of the  
259 specimen before the measurement. It was assumed that the wrap would not exert any constraining  
260 forces on the deformation of the specimen and would allow the liquid water to extrude through the  
261 surfaces of the specimen. The specimen was frozen and thawed in a climatic chamber. The air  
262 temperature in the chamber was maintained at −10 and 20 °C during the cooling and heating  
263 periods, respectively. The duration of the cooling and heating periods was 21 h and 15 h,  
264 respectively, and the rates of temperature change were not controlled. Before the cooling period,  
265 the air temperature in the chamber was kept at 20 °C for one hour to stabilize the temperature of  
266 the specimen.

267



268

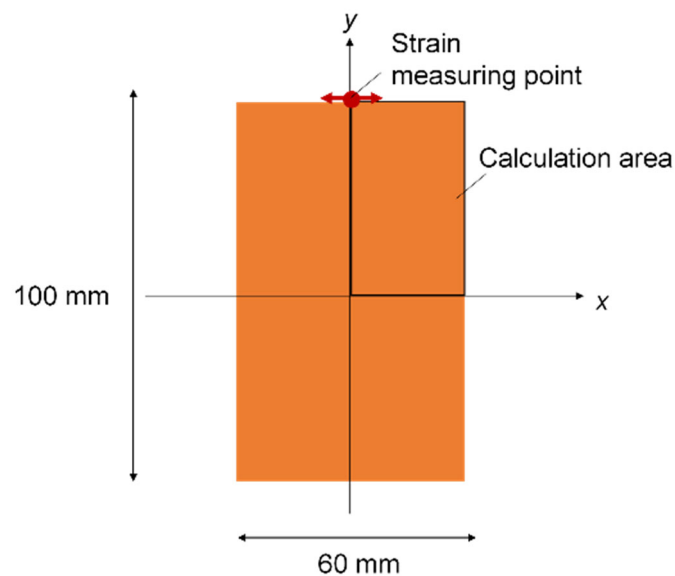
269 **Fig. 4.** Schematic diagram of a brick specimen for strain measurement for model validation

270

### 271 3.3.2. Calculation model and conditions

272 Fig. 5 shows a schematic representation of the calculation model for the strain measurement. The  
273 model corresponded to the 100 mm × 60 mm cross section in the middle of the 210 mm long side of  
274 the brick specimen for the strain measurement shown in Fig. 4. The calculation was performed for a

275 quarter of the cross-section because the specimen was symmetrical. The plane strain condition was  
 276 assumed. We used square elements with 2.5 mm edges and set the time step to 10 s in this calculation.  
 277 The initial temperature in the material was uniform and set to the average of the temperatures measured  
 278 on the top and side surfaces at the beginning of the experiment. One hour before the cooling period  
 279 was included in the calculation. As the specimen was saturated with liquid water at the beginning of  
 280 the experiment, the initial value of  $P_l$  in the material was set to 0 Pa. There was no heat or moisture  
 281 flux on the axes of symmetry ( $x=0$  or  $y=0$  in Fig. 5) and the displacement normal to the axis directions  
 282 was restricted. On the surfaces of the material,  $P_l$  was set to 0 Pa and the heat flow was calculated  
 283 using the Robin boundary condition with the measured air temperature in the test chamber. The heat  
 284 transfer coefficient between the material surfaces and the air in the chamber was determined to be 9.0  
 285  $[W/(m^2 \cdot K)]$  based on the rate of temperature decrease before the onset of freezing in the cooling period.  
 286



287  
 288 **Fig. 5.** Calculation model of a brick specimen for strain measurement for model validation

289  
 290 *3.3.3. Material and water properties*

291 Based on laboratory measurements and the literature, the parameters listed in Table 1-3 were used for  
 292 the calculations. The temperature dependence of the properties was neglected in the calculations  
 293 because it was much less significant than the dependence on water saturation. The water retention

294 curve and thermal conductivity of the material in the unfrozen and unsaturated states were derived  
295 from the literature (Kumaran, 1996) and approximated using van Genuchten's equation (Van  
296 Genuchten, 1980) and a linear function, respectively, as shown in Fig. 6. The parameters in Eq. (9)  
297 were determined as follows:  $\beta = 0.0000357$ ,  $n = 1.52$ , and  $m = 0.342$ . The dependence of thermal  
298 conductivity on ice saturation was estimated using Maxwell's equation, (de Vries, 1963) as in our  
299 previous study (Fukui, et al., 2021):

300

$$301 \quad \lambda = 1.0 + \phi_0 (4.18S_i + 7.15S_i) \quad (17)$$

302

303 **Table 1.** Material properties with a constant value used in the calculations.

Property	Unit	Symbol	Value	Source
Dry density	kg/m <sup>3</sup>	$\rho_d$	1940	Measurement
Water content at vacuum saturation*	m <sup>3</sup> /m <sup>3</sup>	$\phi_0$	0.246	Measurement
Specific heat	J/(kg·K)	$c_d$	920	Literature (Kumaran, 1996)
Vapor permeability**	kg/(m·s·Pa)	$\mu$	$2.6 \times 10^{-11}$	Literature (Kumaran, 1996)
Thermal expansion coefficient	K <sup>-1</sup>	$\alpha$	$3.8 \times 10^{-6}$	Measurement
Young modulus***	Pa	$E$	$19 \times 10^9$	Measurement
Poisson's ratio***		$\nu$	0.32	Measurement
Biot coefficient		$b$	0.57	Eq. (7)

304 \*regarded as the open porosity

305 \*\*used to derive the moisture permeability (Eq. (21))

306 \*\*\*used to calculate the elastic stiffness tensor  $\mathbf{D}$  and bulk modulus  $K$  in Eqs. (1) and (3), respectively.

307

308 **Table 2.** List of hygrothermal and poroelastic parameters used in the calculations, given as a function.

Property	Unit	Symbol	Figure	Equation
Thermal conductivity	W/(m·K)	$\lambda$	Fig. 6 (b)	Eq. (17)
Water retention curve			Fig. 6 (a)	Eq. (9)
Moisture permeability	kg/(m·s·Pa)	$\lambda'$	Fig. 7 (b)	Eq. (18)
Generalized Biot coupling moduli	Pa	$N_{ii}, N_{il}, N_{li}, N_{ll}$		Eq. (4)
Coefficient related to the volumetric thermal expansion of the pore volume	K <sup>-1</sup>	$a_i, a_l$		Eq. (5)

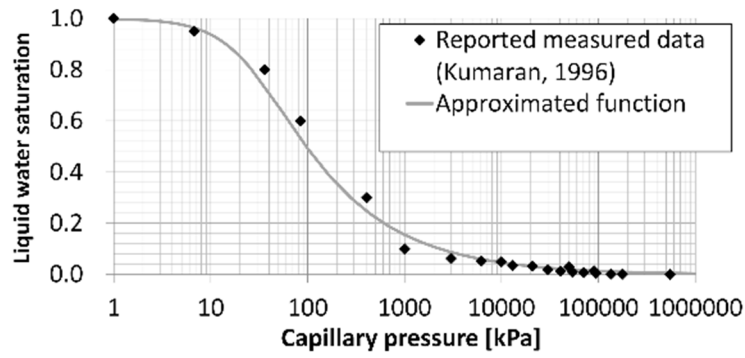
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310

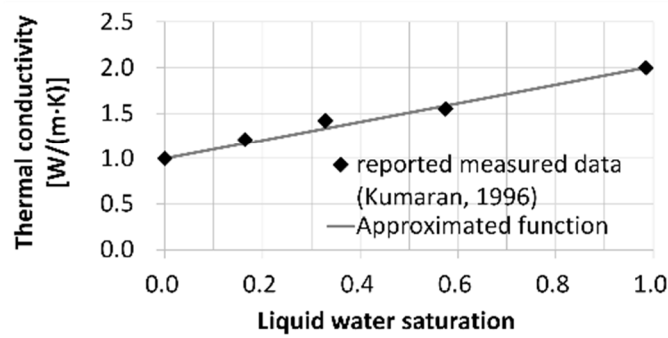
311 **Table 3.** List of properties of water used for calculations.

Property	Unit	Symbol	Value		Source
			Liquid water	ice	
Specific heat	J/(kg·K)	$c$	$4.22 \times 10^3$	$2.117 \times 10^3$	
Bulk modulus	Pa	$K$	$1.79 \times 10^3$	$7.81 \times 10^3$	Literature (Coussy, 2005; Coussy & Monteiro, 2008)
Coefficient of thermal expansion	K <sup>-1</sup>	$\alpha$	$-95.4 \times 10^{-6}$	$51.7 \times 10^{-6}$	Literature (Coussy, 2005; Coussy & Monteiro, 2008)
Density in reference state	kg/m <sup>3</sup>	$\rho^0$	1000	917	

312



(a)



(b)

313

314 **Fig. 6.** (a) Water retention curve and (b) thermal conductivity of a brick in the unfrozen and  
 315 unsaturated state

316

317 The moisture diffusivity derived using the ruler method (Evangelides, et al., 2018; Ren, et al.,  
 318 2019) is shown in Fig. 7 (a). We assumed that the water uptake surface was capillary-saturated during  
 319 the measurement. To extend the obtained function to the vacuum-saturated region, we assumed that  
 320 the moisture diffusivity did not change significantly between the capillary and vacuum-saturated states  
 321 (represented by the dotted line in Fig. 7 (a)). From the moisture diffusivity function and water  
 322 retention curve, the moisture permeability of the unfrozen material was calculated using the  
 323 following equation:

324

$$325 \quad \lambda' = \rho_l \phi_0 D_w \frac{\partial S_l}{\partial P_c} \quad (18)$$

326

327 where  $D_w$  is the moisture diffusivity [ $\text{m}^2/\text{s}$ ] and  $P_c$  is the capillary pressure [Pa]. The derivation of  
328 a moisture transport coefficient using the moisture diffusivity and water retention curve is valid  
329 in the capillary water region (Künzel, 1995). However, in the low liquid water saturation region,  
330 the vapor transport is dominant at normal temperature, which should be omitted in the freezing  
331 processes of a saturated material.

332 The vapor pressure can be expressed as follows (Künzel, 1995):

333

334 
$$P_v = P_{sat} h = P_{sat} \exp\left(\frac{P_c}{\rho_l R_v T}\right) \quad (19)$$

335

336 where  $P_{sat}$  is the saturated vapor pressure [Pa] and  $R_v$  is the gas constant of the vapor [ $\text{J}/(\text{kg}\cdot\text{K})$ ].

337 As  $P_{sat}$  is a function of temperature, the vapor flow  $q_v$  [ $\text{kg}/(\text{m}^2\cdot\text{s})$ ] in a material can be expressed  
338 as follows:

339

340 
$$q_v = -\mu \frac{\partial P_v}{\partial x} = -\mu \left( \frac{\partial P_v}{\partial T} \frac{\partial T}{\partial x} + \frac{\partial P_v}{\partial P_c} \frac{\partial P_c}{\partial x} \right) = -\left( \mu_T \frac{\partial T}{\partial x} + \mu_P \frac{\partial P_c}{\partial x} \right) \quad (20)$$

341

342 where  $P_v$  is the vapor pressure [Pa],  $\mu$  is the vapor permeability of the material [ $\text{kg}/(\text{m}\cdot\text{s}\cdot\text{Pa})$ ], and

343  $\mu_P$  and  $\mu_T$  are the vapor transport coefficients due to capillary pressure and temperature gradient,

344 respectively. To exclude the vapor transfer portion from the total (liquid and vapor) permeability,

345 the following equation was used:

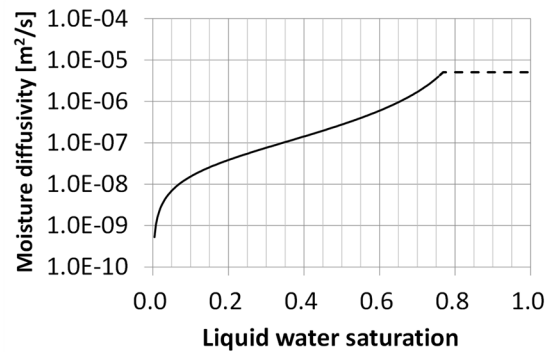
346

347 
$$\lambda'_l = \lambda' - \mu_P = \rho_l \phi_0 D_w \frac{\partial S_l}{\partial P_c} - \mu \frac{P_v}{\rho_l R_v T} \quad (21)$$

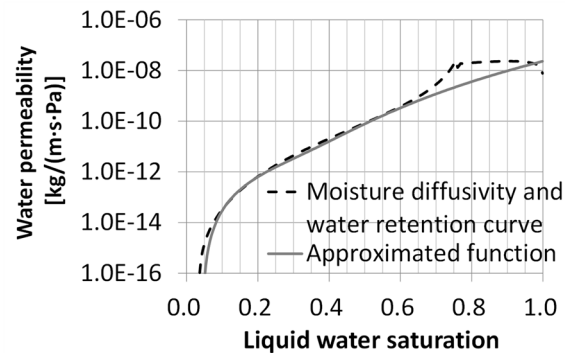
348

349 A vapor permeability of  $2.6 \times 10^{-11}$  [kg/(m·s·Pa)] was derived from the literature (Kumuran, 1996).  
 350 When  $\lambda'$  calculated from Eq. (18) was less than 0,  $\lambda'_v$  was set to 0 [kg/(m·s·Pa)]. The calculated water  
 351 permeability and its approximated function are shown in Fig. 7 (b). We used this approximated  
 352 function for the calculations during freezing and thawing, assuming that moisture permeability is  
 353 similar between the unfrozen and frozen states when the liquid water content was the same (Watanabe  
 354 & Osada, 2016).

355



(a)



(b)

356

357 **Fig. 7.** (a) Moisture diffusivity and (b) water permeability of a brick.

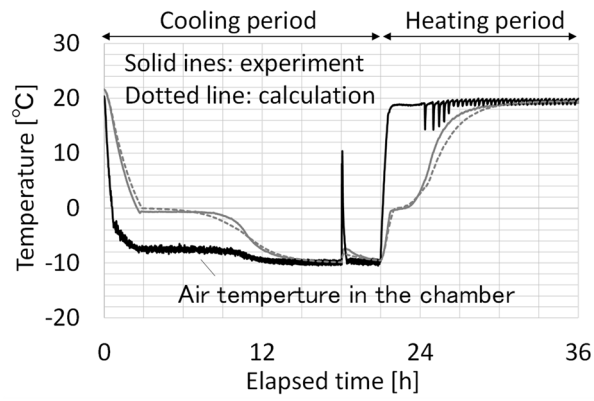
358

### 359 3.3.4. Comparison of measured and calculated results

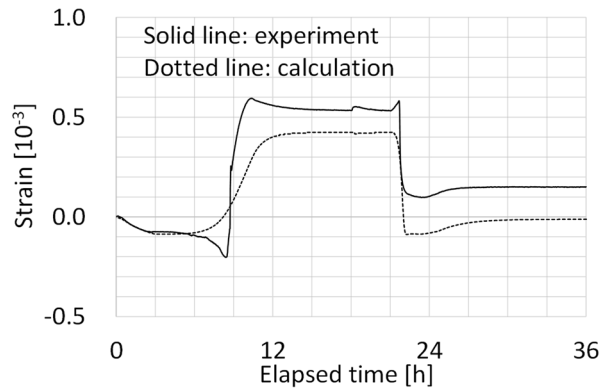
360 Fig. 8 shows a comparison of the measured and calculated temperature and strain evolutions. As shown  
 361 in Fig. 8 (a), the calculated temperature agrees well with the measured values at the surface of the  
 362 specimen. The sudden increase in the measured air temperature in the chamber after 18 h was caused  
 363 by opening the door of the chamber to check the inside. As the anisotropies of the mechanical

364 properties and Biot coefficient were not considered in the calculation, the calculated strain in the  
365 direction normal to the material thickness should be overestimated (Fukui et al., 2023b). However, the  
366 trend of the evolution of the strain was well reproduced in the calculation, except for the decrease after  
367 9 h. The differences in the magnitude of the strain in the calculation and measurement could be due to  
368 the imperfection of material properties, especially moisture permeability. The moisture permeability  
369 of bricks exhibits large variations and is highly dependent on the liquid water saturation (Kumaran,  
370 1996), which can greatly affect magnitude of the strain. In addition, the decrease in measured strain  
371 after 9 h could be due to shrinkage due to the lower pressure of liquid water compared to ice (Eq. 7)  
372 (Sun and Scherer, 2010). In our calculation, the liquid water pressure on the boundary was fixed at 0  
373 Pa. Therefore, the pressure of the ice and liquid water was always positive or equal to 0 Pa, and no  
374 shrinkage could occur on the material surface in the calculation. However, there also should be an ice  
375 phase on the boundary, and it is also possible that the surface had dried slightly after complete  
376 saturation. The treatment of the boundary for the calculation of the freezing processes should be  
377 investigated in the future. Considering the strain as a function of temperature (see Fig. 8 (c)), there is  
378 a relatively large difference between the measured and calculated results. This difference was probably  
379 caused by the three-dimensional effects of heat and moisture transfer and deformation. The strain  
380 remaining after thawing was the residual strain, which cannot be reproduced in our calculation because  
381 plastic deformation was not considered.

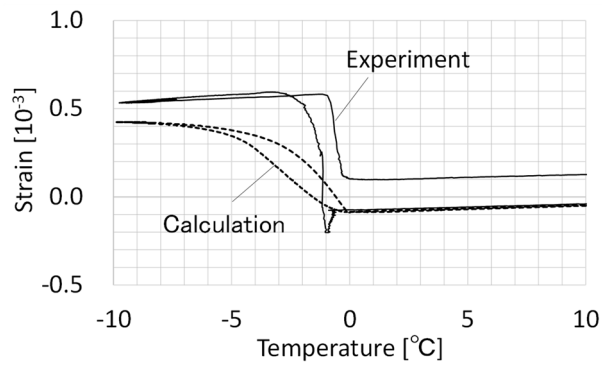
382



(a)



(b)



(c)

383

384 **Fig. 8.** Comparison between measured and calculated results; time evolution of the (a) surface

385 temperature and (b) strain, and (c) strain as a function of temperature

386

## 387 4. Pressure development during cyclic freezing and thawing

### 388 4.1. Calculation methods

#### 389 4.1.1. Calculation models

390 In this section, we describe two calculation cases. Fig. 9 shows the calculation model. The model  
391 corresponded to the cross section of the top brick of the wall, where the cracks were observed, as  
392 shown in Fig. 2 (b). The length of the models was set to 240 mm, which was close to the value given  
393 by Nakamura et al. (2011a; 2016). For case 1, relatively simple conditions were used, which cause  
394 one-dimensional heat and moisture transfer in the material. The two long sides ( $y = 0$  and 5 mm) were  
395 assumed to be insulated and impermeable to moisture, as shown in Fig. 9 (c). One short side ( $x = 240$   
396 mm) was also insulated assuming that the surface was covered with snow, and the other side ( $x = 0$   
397 mm) was exposed to the temperature variations of the outside air (described in the next subsection).  
398 The heat transfer coefficient between the material surface and air was set at  $23 \text{ W}/(\text{m}^2 \cdot \text{K})$ . It was  
399 assumed that the snow does not exert any constraining force on the material and liquid water can move  
400 freely through the two short sides, i.e., the pressure of liquid water was kept at 0 Pa on these sides.  
401 The model consisted of a single sequence of  $5 \times 5$  mm elements, and the time step was set to 10 s.

402 For the calculation case 2, the boundary conditions reflected the actual weather conditions. The  
403 dimensions of the model are  $240 \text{ mm} \times 60 \text{ mm}$ . The top brick of the wall has different dimensions  
404 from common bricks as shown in Fig. 2 (b), but the simplified dimensions were used in the calculation.  
405 As shown in Fig. 9 (d), the conditions on the bottom ( $y = 60 \text{ mm}$ ) and one short side ( $x = 240 \text{ mm}$ )  
406 were the same as in calculation case 1. However, it was assumed that part of the top side ( $y = 0 \text{ mm}$   
407 and  $x > 80 \text{ mm}$ ) was also covered with snow, considering the situation shown in Fig. 2 (b), and the  
408 same boundary conditions as on the right side ( $x = 240 \text{ mm}$ ) were used. On the left part of the top side  
409 ( $y = 0 \text{ mm}$  and  $x < 80 \text{ mm}$ ) and the other short side ( $x = 0 \text{ mm}$ ), solar radiation and air temperature  
410 changes were considered as external conditions. The solar radiation was considered in the form of the  
411 equivalent temperature:

412

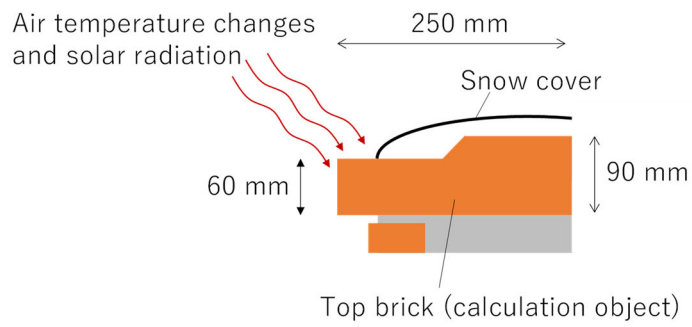
413 
$$T_{eq} = T_o + \frac{rJ}{h} \quad (22)$$

414

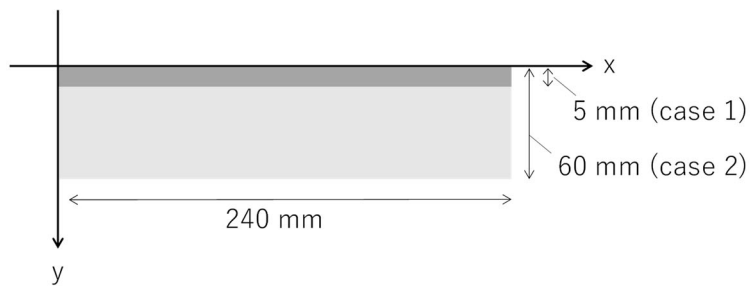
415 where  $h$  is the heat transfer coefficient [ $\text{W}/(\text{m}^2 \cdot \text{K})$ ],  $J$  is the solar radiation on the wall surface [ $\text{W}/\text{m}^2$ ],  
416  $r$  is the solar absorptivity of the material,  $T_{eq}$  is the sol-air temperature, and  $T_o$  is the outside  
417 temperature.  $h$  and  $r$  were set to 23 [ $\text{W}/(\text{m}^2 \cdot \text{K})$ ] and 0.6, respectively. The calculation was performed  
418 using elements with dimensions 20 mm  $\times$  20 mm and a time step of 10 s.

419 For both calculation cases, the initial temperature, liquid water pressure, and material displacement  
420 were set to 0 °C, 0 Pa, and 0 m, respectively. The displacements at the bottom ( $y = 5$ ) for case 1 and  
421 right ( $x = 240$  mm) sides for cases 1 and 2 were constrained in the direction normal to each side.

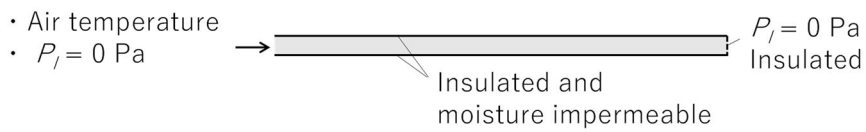
422



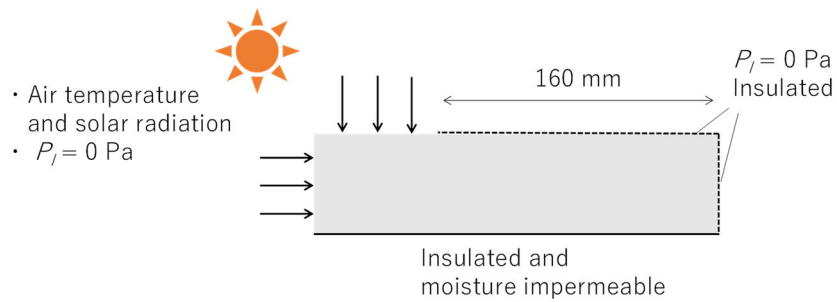
(a)



(b)



(c)



(d)

423

424 **Fig. 9.** Schematic diagram of the calculation object and models; (a) the top brick of the wall, (b)  
 425 simplified dimensions and coordinate system, as well as boundary conditions regarding temperature  
 426 and water pressure for the calculation (c) case 1 and (d) case 2.

427

428 4.1.2. External conditions

429 For the calculation case 1, special temperature condition (see Fig. 10 (a)) was used, which intentionally  
 430 causes the “blockage-type” freezing. During the first cooling period (with an elapsed time of less than  
 431 5.25 d), the material was completely frozen. Subsequently, the external temperature increased and  
 432 reached a maximum of 10 °C during the heating period. At this point, the surface of the material had  
 433 thawed, but the interior remained frozen. At the beginning of the second cooling period (after six days  
 434 had elapsed), the outside temperature dropped below 0 °C, and the material began to freeze again. The  
 435 cooling period was much longer compared with the heating period to ensure complete freezing of the  
 436 material. These temperature changes during the heating period (5.25-6 d) and at the beginning of the  
 437 cooling period (6-6.25 d) were sine curve with a wavelength of 1 d simulating the daily temperature  
 438 fluctuation in a real environment, except for the extra low temperature period (0.25-5.25 and 6.25-12  
 439 d). These periods were added to ensure the complete freezing of the material. The set maximum and  
 440 minimum temperatures of 10 and -10 °C, respectively were possible values in Kitami City, considering  
 441 the air temperature variations in winter (Fig. 1(a)) and temperature increase due to the solar radiation  
 442 during the daytime.

443 The meteorological data provided by the Japan Meteorological Agency (2021) were entered into  
 444 the calculation case 2. Fig. 10 (b) shows the hourly temperature changes in Abashiri City. The  
 445 calculation period was approximately two months, from December 8, 2012 to February 9, 2013, and  
 446 a run-up period of one week was added before this period. Global solar radiation data were divided  
 447 into direct and diffuse elements to calculate vertical and horizontal radiation. First, the radiation during  
 448 the sunny time was expressed using the equations of Bouguer and Berlage as follows:

449

$$450 \quad J_1 = J_0 \tau^{\frac{1}{\sin \theta}} \sin H \quad (23)$$

$$451 \quad J_2 = \frac{1}{2} J_0 \sin \theta \frac{1 - \tau^{\frac{1}{\sin \theta}}}{1 - 1.4 \ln \tau} \quad (24)$$

452

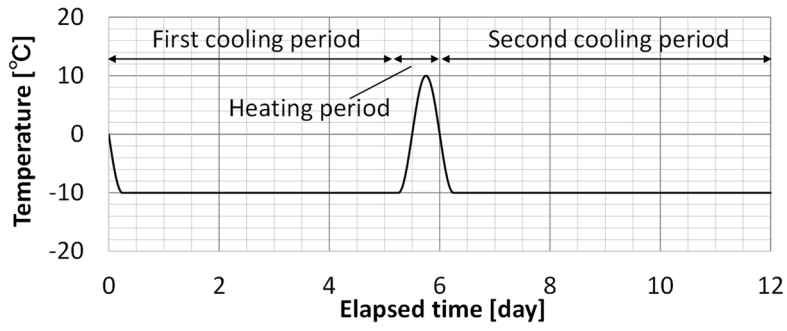
453 where  $H$  is the angle between the direct radiation and the surface of the material;  $J_0$  is the solar constant  
454 [ $\text{W}/\text{m}^2$ ];  $J_1$  and  $J_2$  are the direct and diffuse solar radiation during the sunny periods [ $\text{W}/\text{m}^2$ ],  
455 respectively;  $\tau$  is the atmospheric transmittance; and  $\theta$  is the solar elevation. We set  $\tau$  to 0.79. During  
456 cloudy periods, the total global solar radiation provided was interpreted as diffuse radiation. Finally,  
457 the total radiation was estimated by weighting the solar radiation during sunny and cloudy periods by  
458 the daylight duration  $\chi$  during each hour.

459

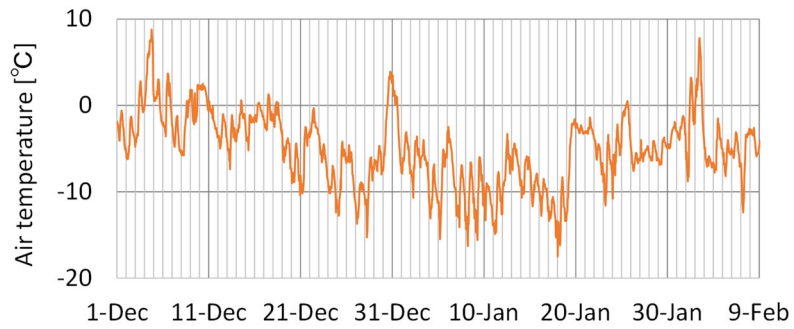
$$460 \quad J = \chi(J_1 + J_2) + (1 - \chi)J_c \quad (25)$$

461

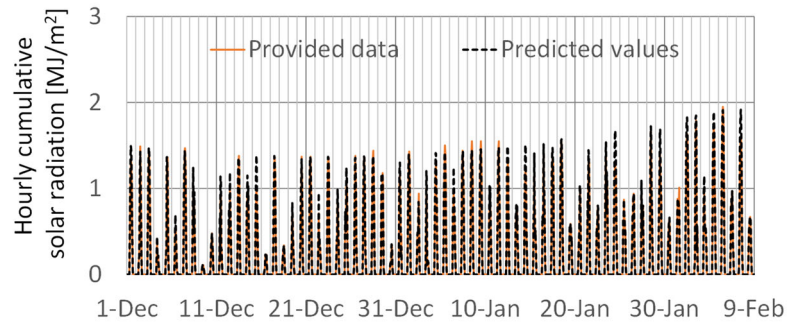
462 where  $J_c$  is the diffuse radiation during the cloudy period [ $\text{W}/\text{m}^2$ ]. Fig. 10 (c) compares the global solar  
463 radiation on horizontal surfaces calculated using Eq. (25) with the meteorological data. Both agree  
464 well, and we calculated the radiation on the vertical surfaces using Eq. (25), as shown in Fig. 10 (d).  
465 In the calculations, we assumed that the vertical surface is oriented to the southeast or northwest. In  
466 these two directions, Nakamura et al. (2011a; 2016) compared the degree of degradation to show the  
467 effects of solar radiation during the day on pressure development due to cyclic freezing and thawing.  
468



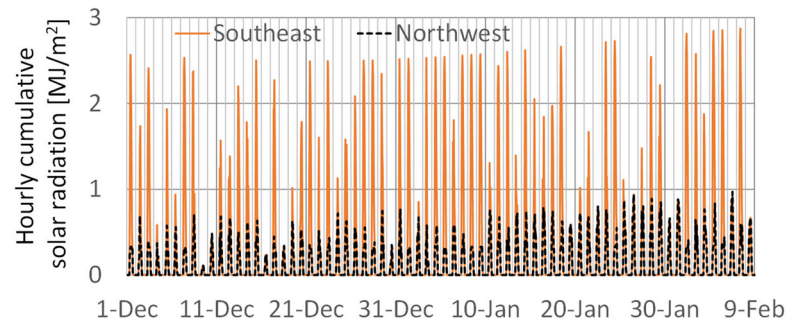
(a)



(b)



(c)



(d)

469

470 **Fig. 10** External conditions used in the calculations: air temperature for (a) case 1 and (b) case 2, and

471 hourly cumulative solar radiation on (c) horizontal and (d) vertical surfaces for case 2.

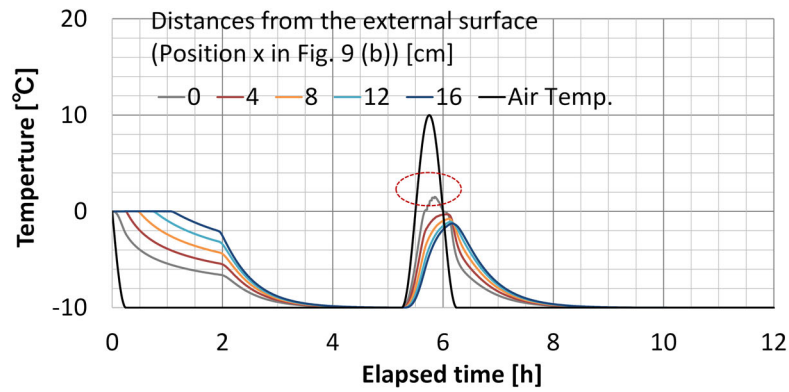
472

## 473 **4.2. Calculation results and Discussion**

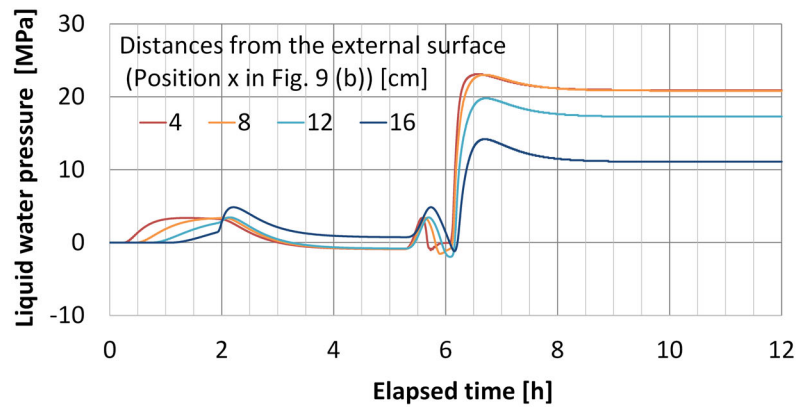
### 474 *4.2.1. Calculation case 1*

475 Figs. 11 and 12 show the time evolution of temperature and liquid water pressure in the material and  
476 ice saturation distribution near the external surface immediately after the beginning of the second  
477 cooling period, respectively. According to Fig. 11 (a), the surface temperature rose above 0 °C during  
478 the heating period (shown with a dotted circle), but the temperature inside the material remained below  
479 0 °C, as intended. Fig. 12 shows that the ice thaws near the surface at the beginning of the second  
480 cooling period (after 6 d) and then gradually refreezes from the surface. This creates an area of low  
481 ice saturation at a distance of 1 to 2 cm from the surface, indicating that the second freeze is the  
482 blocking freeze. Fig. 11 (b) clearly shows that the pressure development in the second cycle was much  
483 stronger than that in the first cycle due to water blockage effects, even though the minimum  
484 temperature of the material was the same in these two cycles. Note that the pressure of the liquid water  
485 sometimes becomes negative because it is always smaller than the pressure of the ice due to the  
486 equilibrium relationship (Eq. (7)). During the first freeze, the pressure at the point where  $x = 16$  cm is  
487 the highest among the other points in Fig. 11 (b). In contrast, during the second freeze, the pressure  
488 tends to be higher in the area near the surface (e.g., at  $x = 4$  and 8 cm) than in the interior of the material.  
489 The high pressure observed near the surface during the second freeze is caused by the low ice  
490 saturation area (at a distance of 1–2 cm from the surface) during the heating period. The water  
491 movement from this area was blocked by the surrounding high ice saturation areas when the water  
492 was frozen again from the material surface during the second freeze.

493



(a)

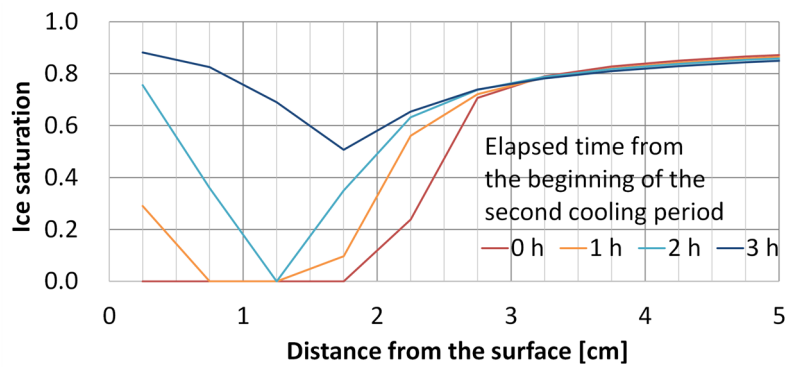


(b)

494

495 **Fig. 11** Calculation results for case 1; evolution of (a) temperature and (b) liquid water pressure

496



497

498 **Fig. 12** Calculation results for case 1; ice saturation distribution immediately after the start of the

499 second cooling period.

500

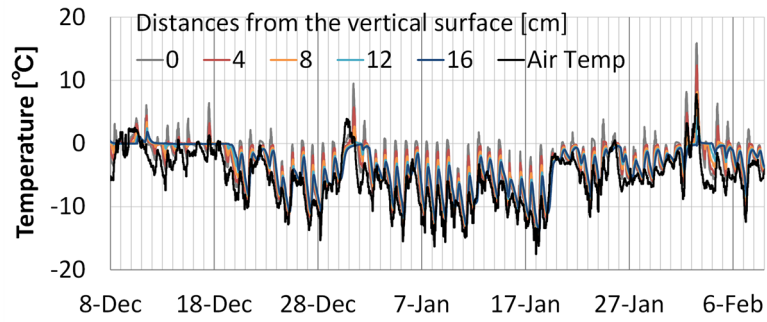
501 4.2.2. Calculation case 2

502 Fig. 13 shows the calculated results of the temperature and water pressure evolution on the bottom  
503 side ( $y = 60$  mm), where the pressure amplitude is the largest. From the comparison of the temperature  
504 evolution of the southeast and northwest facing walls (see Fig. 13 (a) and (b), respectively), it can be  
505 seen that the temperature on the vertical surface ( $x = 0$  mm) of the southeast facing wall often reaches  
506 above  $0$  °C due to solar radiation, whereas such a situation is rare for the northwest facing wall.  
507 Therefore, the surface of the southeast facing wall is exposed to significant freezing and thawing cycles.  
508 Note that the temperature of the inside of the wall seldom rises above  $0$  °C due to its heat capacity,  
509 which means that the inside remains frozen while the surface thaws.

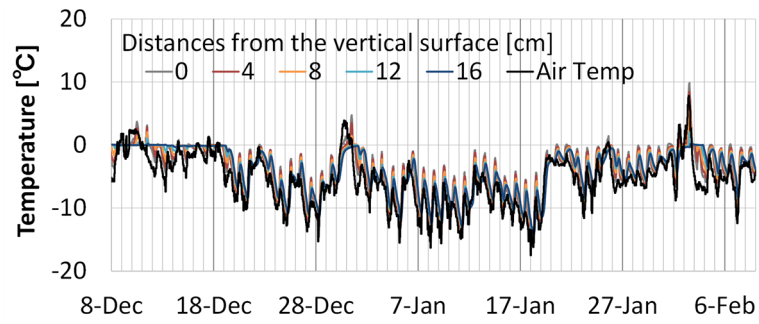
510 These differences in the temperature evolution of the walls oriented to the different directions affect  
511 the pressure development in the walls. As can be seen in Fig. 13 (d), the pressure of the liquid water  
512 increased to 10 MPa or more in the wall facing the north side. Considering that the tensile strength of  
513 conventional bricks is several megapascals, a pressure of this magnitude may cause deterioration of  
514 the walls. However, when the pressure development was observed in the wall facing the south side  
515 (Fig. 13 (c)), the pressure increased more significantly. These results are consistent with the  
516 observations of Nakamura et al. (2011a; 2016) which show that there were tiny cracks on the walls  
517 facing the north side and larger and fatal cracks on the walls facing the south side. It is also worth  
518 noting that the pressure does not reach its highest value in the middle of February, when the  
519 temperature is the lowest. In mid-winter, the temperature cannot be high enough during the day to  
520 thaw a sufficient amount of water near the surface to cause the “blockage-type” freeze.

521 Finally, Fig. 14 shows the evolution of the temperature and pressure around December 23 and 24,  
522 2012, when the pressure reaches the maximum on the southeast facing wall. On December 23, the  
523 temperature near the wall surface was above  $0$  °C during the day and decreased rapidly during the  
524 night. Figs. 15 and 16 show the ice saturation distribution at two time points on December 23 at the  
525 southeast and northwest facing wall, respectively. As shown in Fig. 15, the surface of the material was  
526 completely thawed at 1 pm and there was an ice saturation gradient in the material. As the wall faces  
527 east, it did not receive significant solar radiation in the evening and the surface temperature dropped

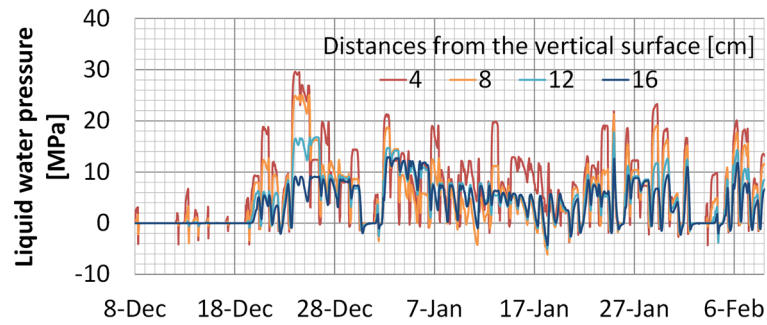
528 rapidly (Fig. 14 (b)). Consequently, the area of low ice saturation is clearly visible at a distance of 30  
529 mm from the material surface at 4 pm, as shown in Fig. 15 (b). Conversely, the saturation in the wall  
530 facing northwest remained high (above 0.75) at 1 pm, and no area of low saturation was formed in the  
531 evening. Such a situation is also found on December 20, 21, 22, and 26 but on December 23 the  
532 temperature increase during the daytime and decrease at night are relatively large. This means that a  
533 large amount of unfrozen water is quickly trapped at night on December 23, resulting in the highest  
534 pressure.  
535



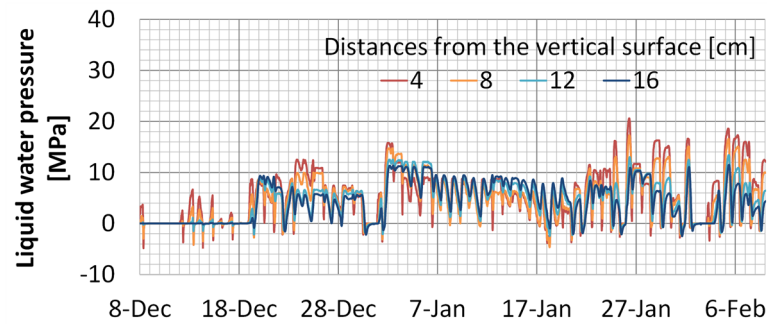
(a)



(b)



(c)



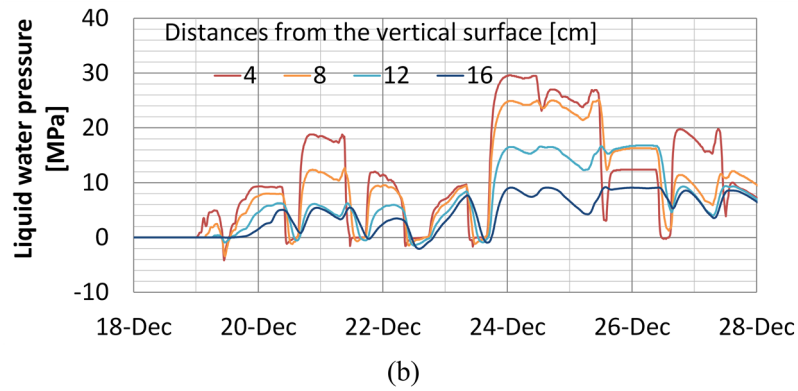
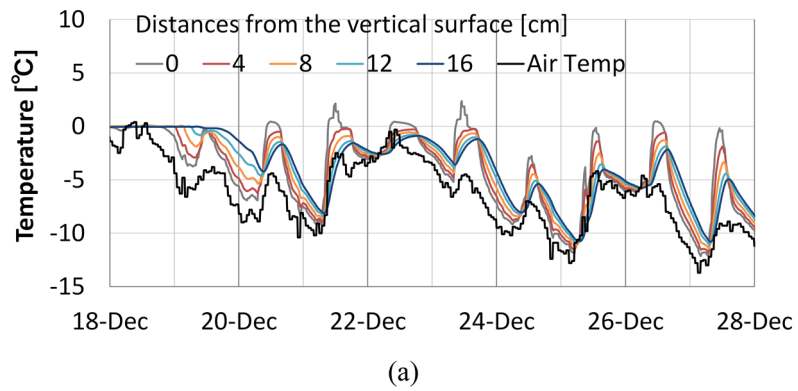
(d)

536

537 **Fig. 13** Calculation results for case 2; temperature evolution in the wall facing (a) southeast and (b)

538 northwest, and the liquid water pressure evolution in the wall facing (c) southeast and (d) northwest.

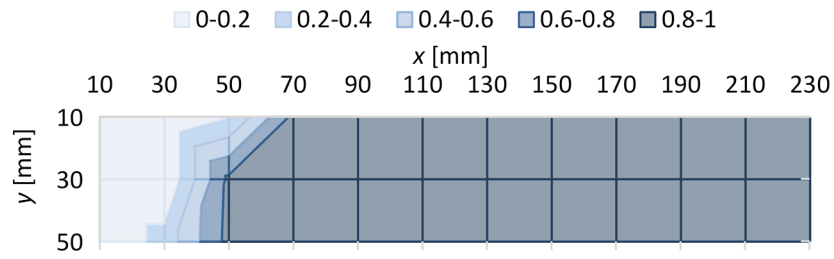
539



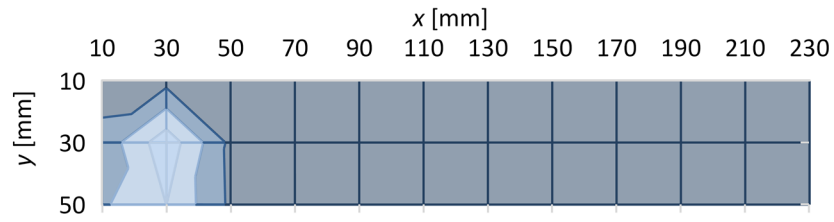
540

541 **Fig. 14** Calculation results for case 2 from December 18 to 28, 2012; evolution of the (a) temperature  
542 and (b) liquid water pressure in the wall facing southeast.

543



(a)



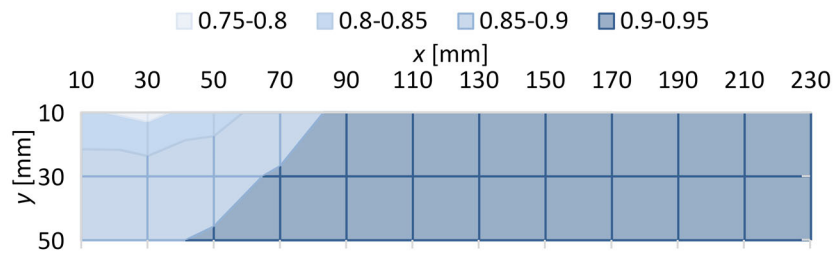
(b)

544

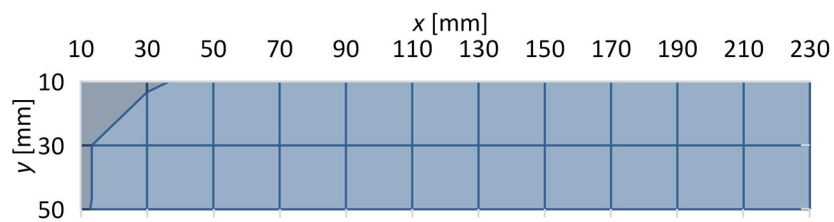
545 **Fig. 15** Calculated ice saturation distribution in the wall facing southeast at (a) 1 pm and (b) 4 pm on

546 December 23.

547



(a)



(b)

548

549 **Fig. 16** Calculated ice saturation distribution in the wall facing northwest at (a) 1 pm and (b) 4 pm on

550 December 23.

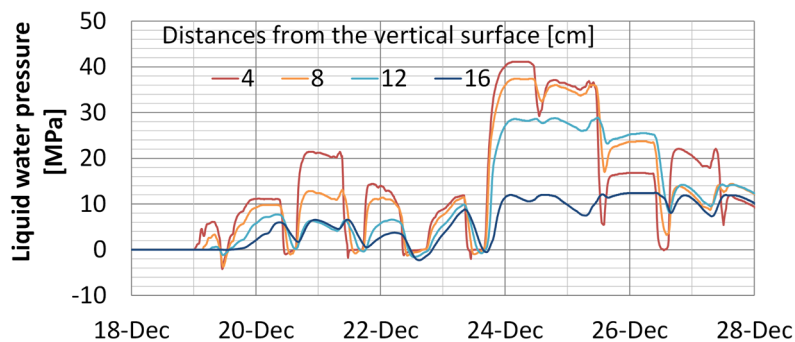
551

552 4.2.3. Influences of moisture permeability on pressure development in the material

553 Of the hygrothermal, mechanical, and poroelastic properties used in the calculations, the transport  
554 coefficient for liquid water exhibits the greatest variation and strongest dependence on liquid water  
555 saturation (Kumaran, 1996). In addition, as water movement in a material plays a key role during the  
556 blocking freeze, the pressure development depends strongly on this property. Therefore, an additional  
557 calculation was performed with a different moisture permeability to discuss its influence on the  
558 pressure development in a material.

559 Fig. 17 shows calculated evolution of the liquid water pressure in the wall facing southeast with  
560 the 1.5 times reduced moisture permeability. From the comparison with Fig. 14 (b), the pressure  
561 development in the night of December 23 was much more significant than in the original case while  
562 the differences in the other periods were smaller. This is because the unfrozen water flow in the  
563 material was more blocked when the moisture permeability of the material was low. The hydraulic  
564 pressure caused by the unfrozen water flow depends on the moisture permeability of a material  
565 (Powers, 1949). The results show that this permeability effect may be more pronounced during the  
566 “blockage-type” freezing.

567



568

569 **Fig. 17** Calculated evolution of the liquid water pressure in the wall facing southeast from December  
570 18 to 28, 2012, when moisture permeability was reduced 1.5 times.

571

572 **5. Conclusion**

573 In this study, we investigated the evolution of pore pressure during cyclic freezing and thawing under  
574 real weather conditions. The hygrothermal and mechanical simulations clearly showed that the water  
575 pressure during a repetition of the freezing and thawing process in which the interior of the material  
576 remains frozen, is significantly higher than when freezing from the completely unfrozen state, as is  
577 common in ordinary freeze-thaw tests. The calculations for brick walls using meteorological data in a  
578 cold region (Kitami City, Hokkaido Prefecture, Japan) showed that for the south facing wall, the  
579 surface repeatedly froze at night and thawed during the day, while the inside hardly thawed. Due to  
580 these characteristic freeze-thaw cycles, the unfrozen water was trapped in the material and much  
581 higher pressure was generated than for the north facing wall. It was also found that the pressure did  
582 not reach its highest value, during the coldest part of the winter, because the temperature was too low  
583 to cause “blockage-type” freezing.

584 Our results showed the detrimental effects of “blockage-type” freezing and indicate that the  
585 unfrozen water movement in a material significantly influences the evolution of pressure. To avoid  
586 the “blockage-type” freeze, it is advantageous for a material to have high water permeability. Materials  
587 with fine pores or large porosity are considered to be also less affected by the “blockage-type” freeze  
588 as they decrease the rate of the freezing at night due to the significant freezing point depression or  
589 large latent heat, respectively.

590 The limitations of this study are probably due to the boundary conditions of the brick walls  
591 considered in this study. Our calculations assumed that the material was covered with snow, which  
592 means that the water supply was sufficient. Therefore, the results of this study can be applied to  
593 building envelopes with continuous water sources, such as rain, snow, or rising damp. In addition, the  
594 constraint forces of the surrounding materials were not considered in the presented calculation. The  
595 cases where a material cannot be considered nearly saturated, e.g., because the water sources are not  
596 regular, such as condensation, or the surface drying is significant, or where the constraint was  
597 considered significant, need to be further investigated, along with modeling the freeze-thaw processes  
598 of unsaturated materials.

599

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604

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