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Stereodivergent Hydrosilylation of 1-Alkynes Catalyzed by $\text{RhI}(\text{PPh}_3)_3$ Leading to (*E*)- and (*Z*)-Alkenylsilanes and the Application to Polymer Synthesis

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Abstract

Hydrosilylation of 1-alkynes catalyzed by $\text{RhI}(\text{PPh}_3)_3$ gives (*E*)- and (*Z*)-alkenylsilanes in a stereodivergent manner by switching the reaction conditions and the order of the addition of the reagents. The reaction is also applied to the synthesis of silicon containing polymers with (*E*) and (*Z*) main-chain structures via polyaddition.

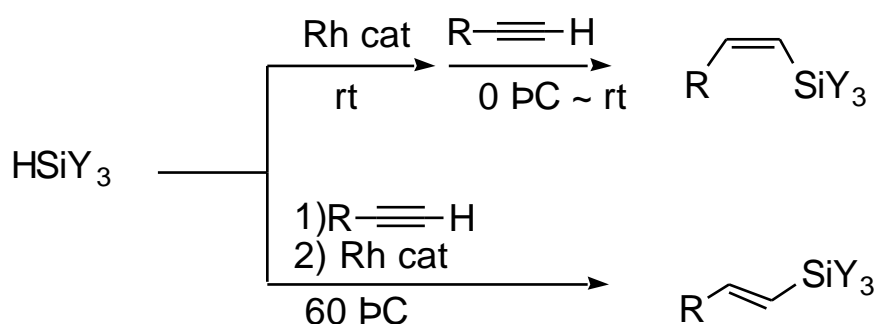
Key words: Hydrosilylation, $\text{RhI}(\text{PPh}_3)_3$, Stereodivergent, Alkenylsilanes, Polyaddition

Hydrosilylation serves as one of the most fundamental reaction in organosilicon chemistry. The reaction proceeds by the catalysis of a variety of transition metals and control of the reactivity and the selectivity in the hydrosilylation have been a major concern in synthetic organic chemistry as well as organometallic chemistry.[1,2] The hydrosilylation of a terminal alkyne with a hydrosilane, for example, would form regio- and stereochemical isomers.

We found that the use of $\text{RhI}(\text{PPh}_3)_3$ as a catalyst for the hydrosilylation of a terminal alkyne furnishes *E* and *Z*-alkenylsilanes in stereochemically divergent manners by switching the order of the addition of the reagents and reaction conditions.[4]

The process is illustrative in Scheme 1. When an alkyne was added to a pretreated mixture of a hydrosilane and the rhodium catalyst at room temperature for 2 h, (*Z*)-alkenylsilane was obtained in a good yield (>99%) with an excellent stereoselectivity (>99:1). On the other hand, reaction of the mixture of a hydrosilane, an alkyne, and the catalyst at 60 °C lead to (*E*)-product (>99% yield, >99% selectivity).

Scheme 1



The results on the stereodivergent hydrosilylation leading to *E*- and *Z*-alkenylsilanes are summarized in Table 1. The hydrosilylation is effective for a variety of hydrosilanes, namely, the ones with heteroatom substituent(s) such as a siloxy or an alkoxy group in contrast that most of the previous rhodium-catalyzed hydrosilylations have been examined with the use of triethylsilane.[1]

Table 1. Stereoselective hydrosilylation of 1-alkynes leading to *Z* and *E*-alkenylsilanes

alkyne	HSiY ₃	condn ^{a)}	temp/°C, time/h	ratio, <i>E</i> : <i>Z</i>
PhC≡CH	HSiMe ₂ OSiMe ₃	A	rt, 2	1:100
		B	60, 1	100:0
	HSiMe(OEt) ₂	A	60, 48 ^{b)}	2:98
		B	60, 1	100:0
4-MeOC ₆ H ₄ C≡CH	HSiMe ₂ OSiMe ₃	A	rt, 2	2:98
4-MeOC ₆ H ₄ C≡CH	HSiMe ₂ OSiMe ₃	B	60, 2	100:0
<i>n</i> -C ₆ H ₁₃ C≡CH	HSiMe(OEt) ₂	A	80, 48 ^{c)}	3:97

a) Cond n A: Pretreatment of a hydrosilane with 0.1 mol% of the rhodium catalyst at rt for 2 h followed by the addition of an alkyne, B: The reaction with a mixture of a hydrosilane, an alkyne and the catalyst.

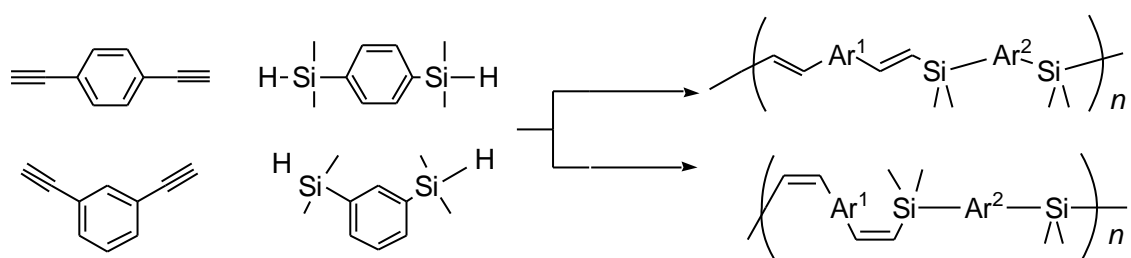
b) The amount of the rhodium catalyst: 0.01 mol%. c) The amount of the rhodium catalyst: 0.005 mol%.

The obtained alkenylsilanes were successfully subjected to transformation reactions of the carbon-silicon bond to a carbon-carbon bond via the palladium-catalyzed cross-coupling reaction with retention of the stereochemistry.[5]

Then, the stereodivergent hydrosilylation is applied to bis-alkynes with bis-hydrosilanes via polyaddition to afford the poly[(phenylene)-((*E*), or

(Z)-vinylene)-(silylene)]s. The reactions proceeded in a highly stereospecific manner as shown in Scheme 2 with the ratios of higher than 90%. The molecular weights and the molecular weight distributions were estimated by SEC analyses ranging 6000-20000 and 2.4-5.5, respectively. UV absorption and emission spectra of the polymers were also measured.[6]

Scheme 2



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