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# Microbial sulfate reduction plays an important role at the initial stage of seafloor sulfide mineralization

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## ABSTRACT

Seafloor hydrothermal deposits form when hydrothermal fluid mixes with ambient seawater, and constituent sulfide minerals are usually interpreted to precipitate abiogenically. Recent research drilling at Izena Hole and Iheya North Knoll in the middle Okinawa Trough (East China Sea), combined with secondary ion mass spectrometry determinations of  $\delta^{34}\text{S}$  in pyrite grains, provides compelling evidence that the initial stage of seafloor sulfide mineralization is closely associated with microbial sulfate reduction. During the sulfide maturation process, pyrite textures progress from framboidal to colloform to euhedral. Pyrite  $\delta^{34}\text{S}$  has highly negative values (as low as  $-38.9\text{‰}$ ) in framboidal pyrite, which systematically increase toward positive values in colloform and euhedral pyrite. Sulfur isotope fractionation between seawater sulfate ( $+21.2\text{‰}$ ) and framboidal pyrite ( $-38.9\text{‰}$ ) is as great as  $-60\text{‰}$ , which can be attained only by microbial sulfate reduction in an open system. Because framboidal pyrite is commonly replaced by chalcopyrite, galena, and sphalerite, framboidal pyrite appears to function as the starting material (nucleus) of other sulfide minerals. We conclude that framboidal pyrite, containing microbially reduced sulfur, plays an important role at the initial stage of seafloor sulfide mineralization.

## INTRODUCTION

Modern seafloor massive sulfide (SMS) deposits are of interest as mineral resources due to their enrichment in  $\text{Cu-Pb-Zn} \pm \text{Au} \pm \text{Ag}$ . Volcanogenic massive sulfide (VMS) deposits are ancient and/or fossil examples of SMS deposits presently observed on land (Ohmoto, 1996; Piercey, 2011); both are formed by seafloor hydrothermal activity related to volcanism in mid-ocean-ridge and arc-back-arc settings (Tivey, 2007; Tornos et al., 2015). Sulfur isotope ( $\delta^{34}\text{S}$ ) analyses of VMS deposits (Lode et al., 2017; Slack et al., 2019; Velasco-Acebes et al., 2019) have shown that an initial mineralization process characterized by the formation of framboidal pyrite (Piercey, 2015) is closely associated with microbial sulfate reduction. However, VMS deposits are subject to diagenesis and metamorphism during their emplacement

ment on land, which can obscure the petrological (mineralogical) and geochemical records of their mineralization process. These early mineralization processes are best studied in SMS deposits, but systematic sampling from beneath the seafloor is possible only by costly deep-sea drilling campaigns.

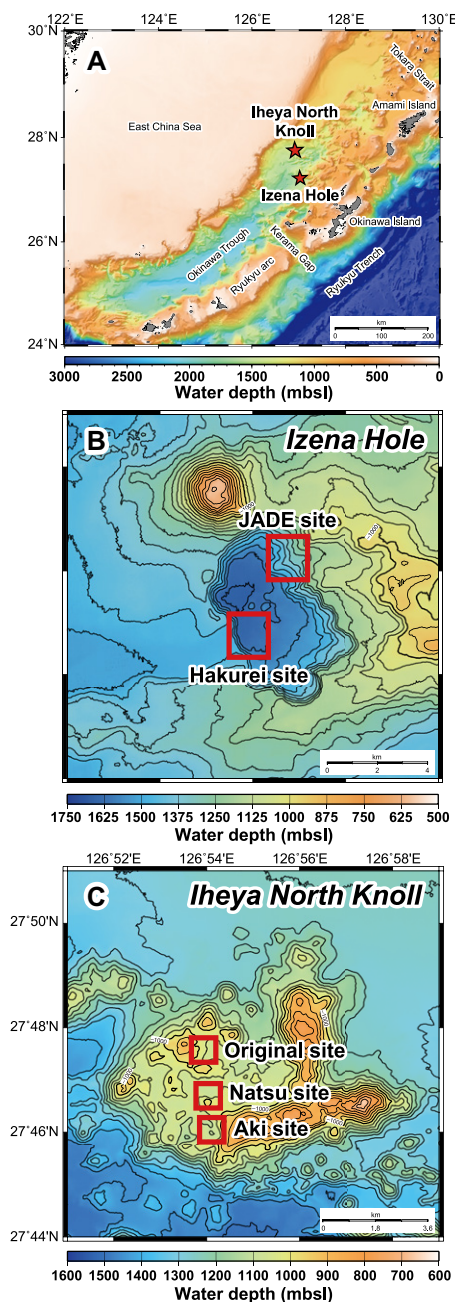
Between 2010 and 2016, D/V *Chikyu* conducted drilling of sulfide deposits at Izena Hole and Iheya North Knoll, middle Okinawa Trough, East China Sea (Fig. 1A), as part of the International Ocean Drilling Program (IODP) Expedition 331 (Takai et al., 2011) and D/V *Chikyu* (JAMSTEC [Japan Agency for Marine-Earth Science and Technology] cruises CK14-04 (D/V *Chikyu* Expedition 907; Takai et al., 2015), CK16-01 (D/V *Chikyu* Expedition 908; Kumagai et al., 2017), and CK16-05 (D/V *Chikyu* Expedition 909; Nozaki et al., 2018). More than

800 m of obtained drill core provided evidence of the mineralization processes, especially the fate of sulfur, in the nascent stage of massive sulfide deposit formation. We also studied pyrite in sulfide chimneys at Iheya North Knoll that formed after the drilling operation at the same drill hole (IODP Hole C0016B) or at another drill hole (Hole C0016A) at the same site where the drill cores were obtained. Our textural observations and determinations of  $\delta^{34}\text{S}$  in this material by *in situ* secondary ion mass spectrometry (SIMS) analysis demonstrate the importance and generality of microbial activity at the seafloor sulfide formation.

## GEOLOGICAL SETTING AND SAMPLES

The Okinawa Trough is a back-arc basin west of the Ryukyu arc in the East China Sea that extends  $>1200$  km from the Japanese mainland to Taiwan (Fig. 1A). Owing to its slow extension rate of  $3.7 \pm 0.06$  cm/yr (Kotake, 2000) and its geomorphological features (Arai et al., 2017), the Okinawa Trough is considered to be at the nascent stage of back-arc basin formation. The basin is divided into northern, middle, and southern segments by the Tokara Strait and Kerama Gap (Fig. 1A).

Our samples are drill cores from Izena Hole and cores and chimneys from Iheya North Knoll in the middle Okinawa Trough (Fig. 1A). Izena Hole (Fig. 1B) has two hydrothermal sites: the JADE site on the northeastern caldera slope (Halbach et al., 1989; Ishibashi et al., 2015) and the Hakurei site on the southern caldera floor (Ishibashi et al., 2015; Totsuka et al., 2019; Morozumi



**Figure 1. (A) Map of the Okinawa Trough, East China Sea. (B,C) Bathymetric maps of Izena Hole and Iheya North Knoll showing hydrothermal site locations. Modified from Nakamura et al. (2015). mbsl—meters below sea level.**

et al., 2020). Iheya North Knoll (Fig. 1C) has three such sites: the Original, Natsu, and Aki sites (Nakamura et al., 2015). All five sites are active hydrothermal sites where the temperature of the hydrothermal fluid effluence exceeds 300 °C. The tectonics, geology, mineralization, and geochemistry of each hydrothermal field were reviewed by Ishibashi et al. (2015).

At Izena Hole, cores were obtained during D/V *Chikyu* Expedition 909 at the Hakurei site, from Hole C9027B on Northern mound and from Hole C9025A east of Northern mound and associated with a seafloor sulfide body

beneath the sediment (Figs. S1 and S2 and Tables S1–S3 in the Supplemental Material<sup>1</sup>) (Nozaki et al., 2018; Totsuka et al., 2019). At Iheya North Knoll, cores were drilled during IODP Expedition 331 at the Original site at Hole C0016B on the eastern flank of North Big Chimney mound (Figs. S1 and S3; Tables S1–S3) (Takai et al., 2011; Yeats et al., 2017). Young and infant chimneys at Holes C0016A and C0016B formed after the drilling operation at artificial hydrothermal vents (i.e., drill holes releasing hydrothermal fluid). Samples from these chimneys, which allow secular changes of mineral precipitation (chimney formation) and hydrothermal fluid geochemistry to be observed and *in situ* experiments to be conducted (Takai et al., 2011; Kawagucci et al., 2013), were collected 5, 11, and 18 months after the IODP Expedition 331 drilling during JAMSTEC research cruises KY11-02 Leg 3 (R/V *Kaiyo*), NT11-16 (R/V *Natsushima*), and NT12-06 (R/V *Natsushima*), respectively. Their petrographic and geochemical details are given by Nozaki et al. (2016).

Under the microscope, pyrite grains from the Hakurei site exhibit various textures (Fig. 2). In pumice fragments and in a sulfidic vein in the hanging wall above the seafloor sulfide body, most pyrite has a framboidal texture and is accompanied by pyrrhotite pseudomorphs replaced by pyrite and marcasite (Fig. 2A). The framboidal pyrite is commonly overgrown by colloform and euhedral pyrite associated with sulfide maturation (Figs. 2B and 2C). This progression from framboidal to colloform and euhedral texture has also been reported in VMS deposits (Piercey, 2015; Velasco-Acebes et al., 2019). In drill cores from the Original site, paragenesis of framboidal and euhedral pyrite is observed in thin section (Fig. 2D), where framboidal pyrite is commonly replaced by chalcopyrite and galena, indicating that framboidal pyrite formed at an initial stage of sulfide mineralization (Halbach et al., 1993; Piercey, 2015). Chimneys from the Original site exhibit two distinct pyrite textures, colloform and/or spherical on the outer side and euhedral and/or acicular on the inner side (Figs. 2E and 2F).

## ANALYTICAL METHODS

*In situ*  $\delta^{34}\text{S}$  analysis was performed by SIMS (CAMECA IMS1280-HR) at the Kochi Institute for Core Sample Research, JAMSTEC (Nankoku, Japan). A primary  $^{133}\text{Cs}^+$  ion beam with an intensity of 100 pA and total impact energy of 20 kV was focused to 2–3  $\mu\text{m}$  in diameter at the sample surface. A gold coat ~30 nm thick was applied to the sample surface, and a normal-incidence electron gun was used for charge compensation. Sec-

ondary ions of two sulfur isotopes were accelerated at 10 kV and detected simultaneously by two Faraday cup (FC) detectors with  $10^{11} \Omega$  amplifiers (L1 for  $^{32}\text{S}^-$  and optically axial FC2 for  $^{34}\text{S}^-$ ). The entrance slit width was set at 61  $\mu\text{m}$ . The exit slit widths were set at 500 and 243  $\mu\text{m}$  for L1 and FC2, corresponding to a mass-resolving power of ~2200 and ~5000, respectively.

Each analysis consisted of 30 s for pre-sputtering, 50 s for centering of secondary ions in the field aperture, and 40 s for measurement (20 cycles with 2 s integration time). The typical intensity of  $^{32}\text{S}^-$  was  $\sim 7 \times 10^7$  counts per second (cps) for pyrite analyses. Each group of nine to 21 sample analyses was bracketed by ten to 12 analyses of pyrite standard UWPY-1 (Ushikubo et al., 2014). Instrumental mass bias and analytical error were determined from average values and two standard deviations (SD) of the bracketing standard analyses. All data were normalized to the Vienna Canyon Diablo troilite (VCDT) standard. Typical uncertainty of  $\delta^{34}\text{S}$  was  $\pm 0.4\text{‰}$  (2SD). Detailed analytical and data reduction procedures were given by Ushikubo et al. (2014).

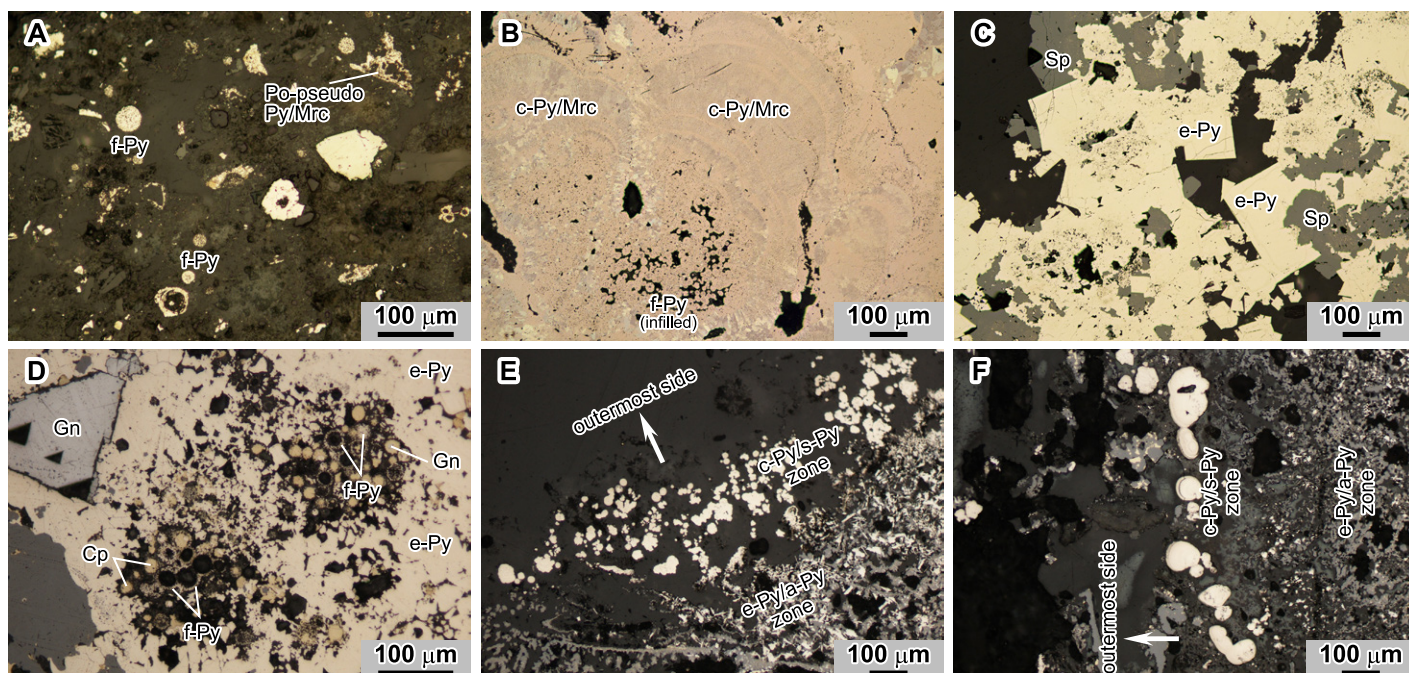
## RESULTS AND DISCUSSION

The SIMS measurements yielded a total of 182  $\delta^{34}\text{S}$  values, representative examples of which are shown in Figure 3 (all measurement sites and data are shown in Figs. S4–S7 and listed in Tables S1–S3). Histograms of the  $\delta^{34}\text{S}$  data, classified by pyrite texture, are shown in Figure 4. In the drill cores from the Hakurei site,  $\delta^{34}\text{S}$  of framboidal pyrite ranged widely from  $-38.9\text{‰}$  to  $-7.1\text{‰}$  (average  $\pm 1\text{SD}$ :  $-17.3\text{‰} \pm 10.2\text{‰}$ ;  $n = 42$ ) except in five analyses of highly recrystallized material. The colloform pyrite had a higher  $\delta^{34}\text{S}$  range, from  $-13.6\text{‰}$  to  $-3.0\text{‰}$  ( $-7.4\text{‰} \pm 2.5\text{‰}$ ;  $n = 29$ ); similarly,  $\delta^{34}\text{S}$  of euhedral pyrite ranged from  $-13.4\text{‰}$  to  $-3.8\text{‰}$  ( $-6.8\text{‰} \pm 2.7\text{‰}$ ;  $n = 19$ ), with one outlier. Recrystallized framboidal pyrite that consisted of several amalgamated grains (infilled framboidal pyrite; Wei et al., 2016) had  $\delta^{34}\text{S}$  of  $\sim 10\text{‰}$  (Fig. 3B), similar to  $\delta^{34}\text{S}$  of colloform and euhedral pyrites, whereas framboidal pyrite retaining its initial texture (normal framboidal pyrite; Wei et al., 2016) in the hanging-wall pumice fragments and sulfidic vein had the lowest  $\delta^{34}\text{S}$  of  $\sim -35\text{‰}$  (Fig. 3A; Fig. S5). Similarly, in drill cores from the Original site, framboidal pyrite had lower  $\delta^{34}\text{S}$  than euhedral pyrite (Figs. 3C and 4; Fig. S4). Unlike in the drill cores, however, pyrites in chimneys at this site had a narrow  $\delta^{34}\text{S}$  range around  $0\text{‰}$ , irrespective of their texture (Figs. 3D and 4; Fig. S7).

Given the  $\delta^{34}\text{S}$  value of  $+21.2\text{‰}$  in bottom-seawater sulfate at Iheya North Knoll (Aoyama et al., 2014), the sulfur isotope fractionation between seawater sulfate and framboidal pyrite was as great as  $-60\text{‰}$ , assuming the same  $\delta^{34}\text{S}$  in seawater sulfate at Iheya North Knoll and Izena Hole. This fractionation factor is close

<sup>1</sup>Supplemental Material. Supplemental figures and tables. Please visit <https://doi.org/10.1130/GEOLOGY.12964949> to access the supplemental material, and contact [editing@geosociety.org](mailto:editing@geosociety.org) with any questions.





**Figure 2.** Reflected-light photomicrographs showing mineralogy and pyrite textures. (A) Framboidal pyrite and pyrrhotite pseudomorph replaced by pyrite and marcasite in sulfidic vein in a hanging wall above the subseafloor sulfide body from the Hakurei site (Okinawa Trough; sample 909-C9025A-3H-7-W, 58.0–60.0 cm [see Fig. S1; see footnote 1]). (B) Infilled framboidal pyrite and colloform pyrite and/or marcasite in subseafloor sulfide body from the Hakurei site (sample 909-C9025A-5H-3-W, 0.0–2.0 cm). (C) Euhedral pyrite associated with sphalerite at the Northern mound, Hakurei site (sample 909-C9027B-2X-CC-W, 15.0–17.0 cm). (D) Framboidal and euhedral pyrite from the eastern flank of North Big Chimney mound, Original site (Yeats et al., 2017) (sample 331-C0016B-1L-1-W, 17.0–19.0 cm). (E,F) Colloform and/or spherical pyrite on the outer side and euhedral and/or acicular pyrite on the inner side of chimneys at IODP Hole C0016A (sample HPD1355R03) (E) and Hole C0016B (sample HPD1247G01) (F), Original site (Nozaki et al., 2016). a—acicular; c—colloform; Cp—chalcopyrite; e—euhedral; f—framboidal; Gn—galena; Mrc—marcasite; Po—pyrrhotite; pseudo—pseudomorph; Py—pyrite; s—spherical; Sp—sphalerite.

to the largest values recorded in VMS deposits ( $-73\text{‰}$  [Lode et al., 2017],  $-71\text{‰}$  [Slack et al., 2019],  $-66\text{‰}$  [Velasco-Acebes et al., 2019]). Three dominant sulfur sources for pyrite in SMS deposits have been considered: (1) magmatic sulfur in footwall rocks leached by hydrothermal fluid through water-rock interaction ( $\delta^{34}\text{S} = 0\text{‰} \pm 3\text{‰}$ ; Campbell and Larson, 1998; Shanks, 2001), (2) thermochemical (abiotic) sulfate reduction of anhydrite and/or gypsum and seawater sulfate by  $\text{Fe}^{2+}$ -bearing minerals or organic carbon, and (3) microbial sulfate reduction of seawater sulfate. A solely magmatic sulfur source to explain the highly negative  $\delta^{34}\text{S}$  in framboidal pyrite can be ruled out at Izena Hole and Iheya North Knoll.  $\delta^{34}\text{S}$  in abiotic  $\text{H}_2\text{S}$  depends on the degree of sulfate reduction, temperature, and  $\delta^{34}\text{S}$  of S in anhydrite, gypsum, and seawater. The equilibrium fractionation factor between  $\text{SO}_4^{2-}$  and  $\text{H}_2\text{S}$  is estimated to be  $\sim 30\text{‰}$  at  $200^\circ\text{C}$ ,  $\sim 25\text{‰}$  at  $250^\circ\text{C}$ , and  $\sim 20\text{‰}$  at  $350^\circ\text{C}$ , and abiotic sulfate reduction is kinetically inhibited below  $200^\circ\text{C}$  (Ohmoto and Lasaga, 1982; Ohmoto, 1996). Because  $\delta^{34}\text{S}$  in sulfate minerals (anhydrite and barite) at Izena Hole and Iheya North Knoll ranges from  $+16.3\text{‰}$  to  $+25.5\text{‰}$  (Zeng et al., 2000; Lüders et al., 2001; Ueno et al., 2003) and  $\delta^{34}\text{S}$  of bottom-seawater sulfate is  $+21.2\text{‰}$  (Aoyama et al., 2014), the  $\delta^{34}\text{S}$  of  $-38.9\text{‰}$  in framboidal pyrite could not have been produced abiotically. Thus, the

strongly negative  $\delta^{34}\text{S}$  in framboidal pyrite could have been produced only by microbial reduction of seawater sulfate. A repeated cycle of reduction of seawater sulfate to sulfide followed by sulfide oxidation (Canfield, 2001) in an open system with excess sulfate is the most plausible process to explain the extremely negative  $\delta^{34}\text{S}$  in framboidal pyrite. An alternative process is single-step microbial sulfate reduction in an open system without sulfur disproportionation and reoxidation (Sim et al., 2011).

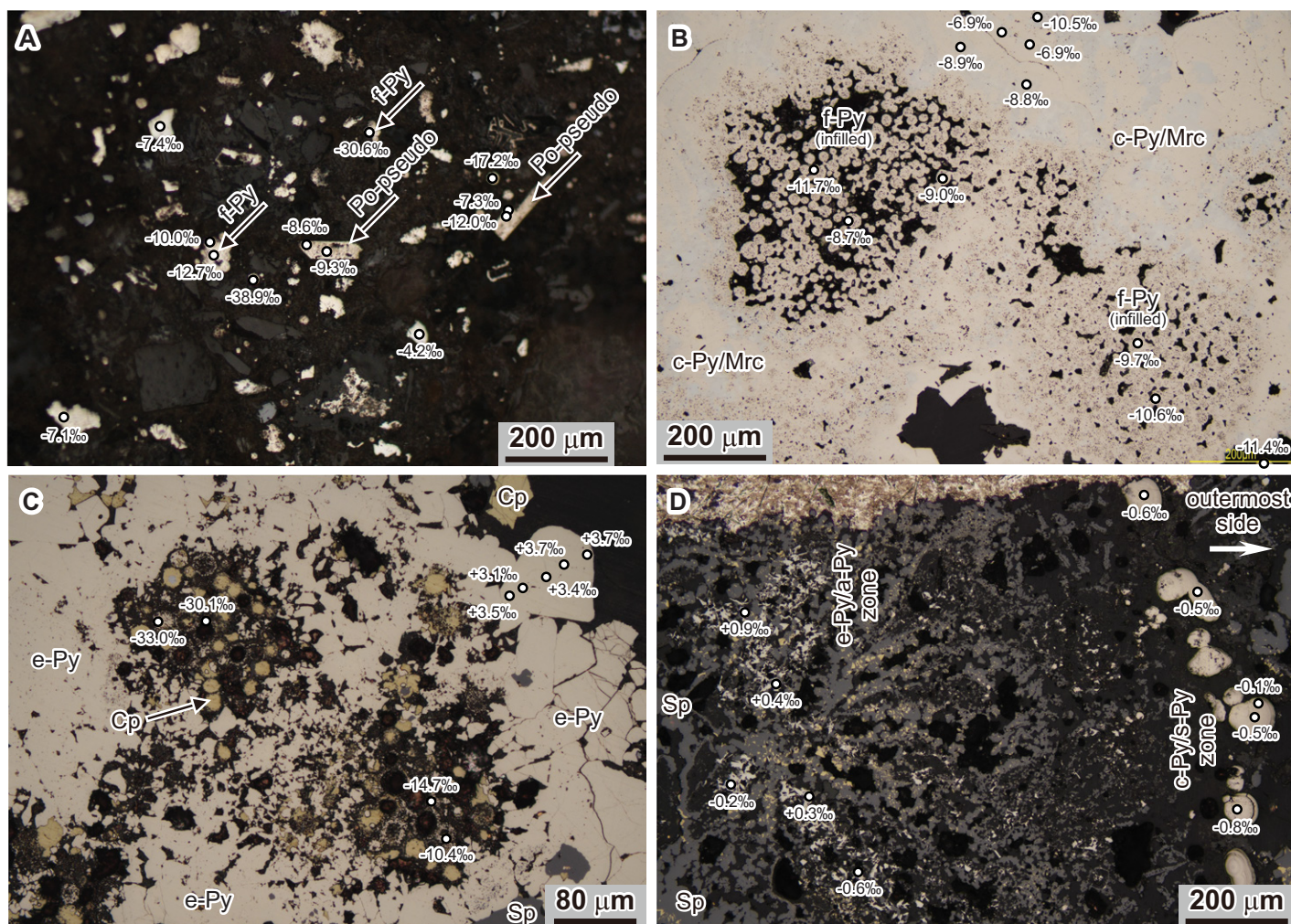
That framboidal pyrite is commonly replaced by other sulfide minerals such as chalcopyrite and galena (Fig. 2D) as well as sphalerite (Piercey, 2015) indicates that framboidal pyrite provides component material for other sulfide minerals. Our evidence suggests that framboidal pyrite incorporating S derived from microbial sulfate reduction serves as a nucleus for subsequent mineral growth at the initial stage of subseafloor sulfide mineralization. Colloform pyrite with  $\delta^{34}\text{S} \sim -7\text{‰}$  is overgrown on framboidal pyrite as a result of overprinting by hydrothermal mineralization. We consider the S in colloform pyrite to be a mixture of magmatic S from rocks, as well as S from abiotic sulfate reduction of anhydrite and/or gypsum and seawater sulfate, with microbially reduced seawater sulfate. Euhedral pyrite in drill cores and chimneys has  $\delta^{34}\text{S}$  of  $0\text{‰}$ – $3\text{‰}$ , explained by further overprinting by hydrothermal mineral-

ization (and possibly additional S input through abiotic reduction). The interstitial water geochemistry of drill cores from the Hakurei site shows positive peaks of alkalinity and  $\text{NH}_4$ ,  $\text{H}_2\text{S}$ , and  $\text{CH}_4$  concentrations at 5–10 m above the subseafloor sulfide body, indicating that vigorous microbial sulfate reduction is taking place within the hanging-wall (pumiceous) sediment. Active subseafloor microbial sulfate reduction has also been confirmed by a multiple sulfur isotope study at the Original site (Aoyama et al., 2014) as well as at the Palinuro and Panarea hydrothermal fields in the Tyrrhenian Sea (Peters et al., 2011). The maximum fractionation in  $\delta^{34}\text{S}$  between framboidal pyrite and seawater sulfate ( $-60\text{‰}$ ) together with  $\delta^{34}\text{S}$  evolution through the sulfide maturation process, observed in SMS and in VMS deposits on land (Lode et al., 2017; Slack et al., 2019; Velasco-Acebes et al., 2019), suggest that microbial activity is essential for SMS and VMS deposit formation and that microbial sulfate reduction plays an important role at the initial stage of mineralization beneath the seafloor sediments.

## CONCLUSIONS

Microscopic evidence and  $\delta^{34}\text{S}$  values in pyrite from Izena Hole and Iheya North Knoll indicate that pyrite progresses through framboidal, colloform, and euhedral textures accompanied by changes in  $\delta^{34}\text{S}$  to higher values. Because





**Figure 3.** Reflected-light photomicrographs showing representative  $\delta^{34}\text{S}$  values in pyrite determined by secondary ion mass spectrometry (SIMS). (A) Sample 909-C9025A-3H-7-W, 58.0–60.0 cm (see Fig. S1 [see footnote 1]). (B) Sample 909-C9025A-5H-3-W, 0.0–2.0 cm. (C) Sample 331-C0016B-1L-1-W, 17.0–19.0 cm. (D) Sample HPD1247G01. Abbreviations as in Figure 2.

$\delta^{34}\text{S}$  in framboidal pyrite as negative as  $-38.9\text{‰}$  (or  $-60\text{‰}$  with respect to seawater sulfate) could not have been produced by magmatic sulfur or abiotic sulfate reduction, the framboidal pyrite must be derived from seawater sulfate through microbial reduction in an open system. The ubiquity and abundance of framboidal pyrite with strongly negative  $\delta^{34}\text{S}$  at the initial stage of sulfide mineralization in both SMS and VMS deposits can thus be explained by microbial sulfate reduction, and its widespread replacement by other sulfide minerals indicates that framboidal pyrite serves as a nucleus for subsequent mineral growth.

#### ACKNOWLEDGMENTS

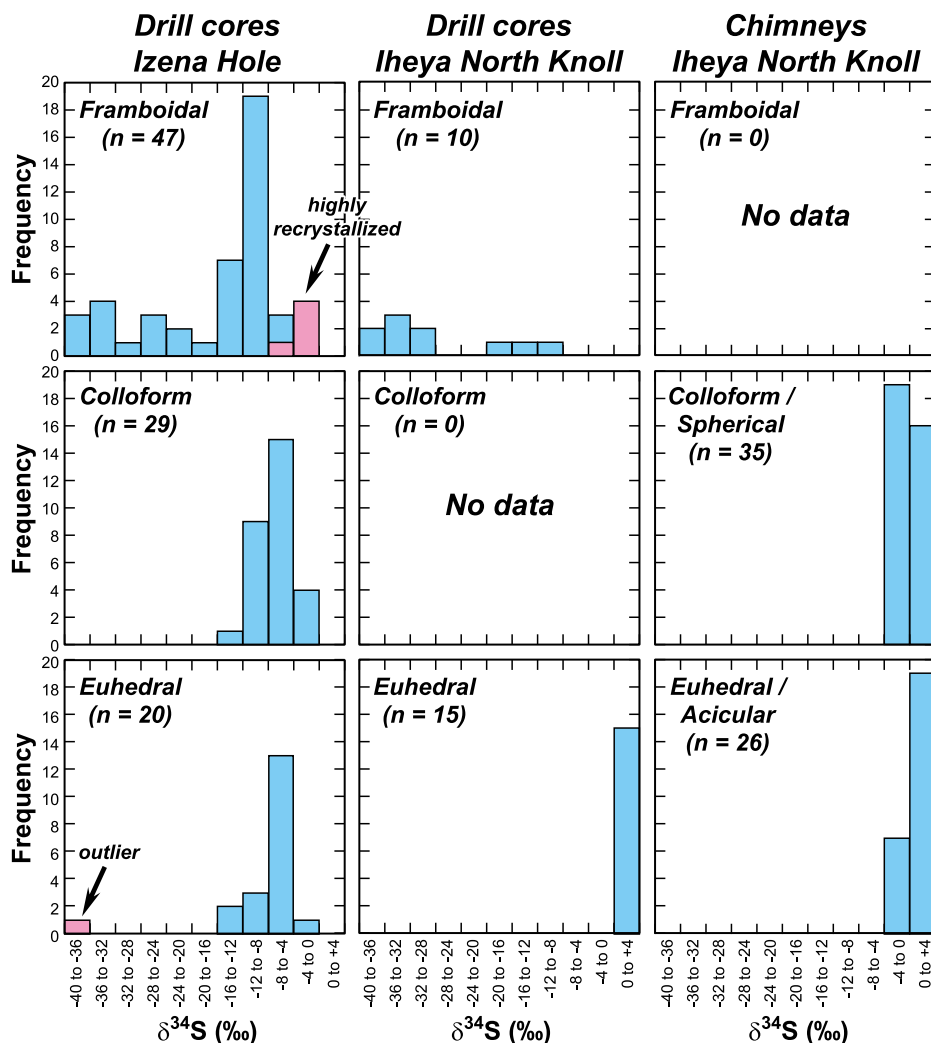
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**Figure 4. Histograms of  $\delta^{34}\text{S}$  in pyrite textural categories. Pink bars in framboidal and euheedral pyrite of Izena Hole represent highly recrystallized framboidal pyrite values and outlier value, respectively.**

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